

The Biogenic Hypothesis: Microbial Acids and Gas as an Explanation for the Dissolution and Forming of Pores and Caves in Limestone*

Stephanie J. Schwabe¹, Rodney A. Herbert², and James L. Carew³

Search and Discovery Article #50385 (2011)

Posted February 25, 2011

*Adapted from oral presentation at AAPG Eastern Section Meeting, Kalamazoo, Michigan, September 25-29, 2010. Abstracts of meeting posted as Search and Discovery Article #90116 (2010)

¹University of Kentucky, Earth & Environmental Sciences, Lexington, KY 40506 (Steffi@blueholes.org)

²University of Dundee, Dundee, Scotland

³College of Charleston, Charleston, SC 29424

Abstract

The current explanation for the formation of pores, holes and caves in marine limestone is the combination of acidic meteoric water and mixing of CaCO₃-supersaturated seawater with CaCO₃-saturated groundwater creating under saturated mixed water. We now know that the mixing of these bodies of water cannot dissolve limestone unless there is a continuous source of acids to overcome the buffering effects of the limestone. Similarly, rainwater that is slightly acidic via incorporation of atmospheric and soil CO₂ cannot remain acidic during its passage through marine-limestone.

Large numbers of interstitial bacteria are the obvious sources of acid and gas needed to initiate and maintain acidic meteoric and ground-water conditions. We have shown that rainwater over San Salvador Island before contact with the ground has a pH of 5.6 and is buffered within three minutes to 8.6 following contact with local limestone. Buffered rainwater passing through 3 meters of limestone arrives in vadose cave as acidified drip-water with a pH of 6.7. We hypothesize that the dissolution capacity of meteoric-water is controlled by the assimilation of bacterially generated acids and gas, produced in the pores of the host rock. Heterotrophic bacteria also produce large quantities of gas in unconsolidated rock and can generate blisters or pores. The gas they generate is temporarily trapped in their mucilage causing a ballooning effect and subsequent displacement of the surrounding unconsolidated grains. The resulting sedimentary textures formed by microbial acid and gas, in modern carbonate rocks, appear to match textures observed in ancient carbonate cores.

Selected Reference

Garrison, B.V., and B.E. Urban, 2003, The effects of OCTG (Oil Country Tubular Goods) connection swaging and stress relieving on SSC (Sulfide Stress Cracking) resistance : NACE International Corrosion Conference, San Diego, California, March 16-20, 2003, paper 03111, 18 p.

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Making holes in rocks via rain water

The solution of CO_2 (g) in water is as follows:



If carbonic acid dissociates:



Abiotic approach



Presenter's Notes: Additional CO_2 is picked up in the soil (no comment as to the origin of the CO_2); this carbonic acid solution now is supposed to make its way through meters of rock and dissolve massive caves?? IF so effective, what about the rock between the surface and the water table? Dissolution stops a few meters from the top of the rock and picks up again at depth??



What is actually observed

pH of rain = 5.7

Soils are absent or poorly developed in the Bahamas

Rain after 3 minutes contact with limestone, pH = 8.6

Limestone rain catchment after one hour: pH = 10.0!

Vadose Zone

pH of drip water coming right out of the stalactite = 6.8
 (NB: the drip water may have a residency time of about 2 to 2.5 months)

pH of drip water pool directly below drip = 7.3-6.9

Phreatic Zone

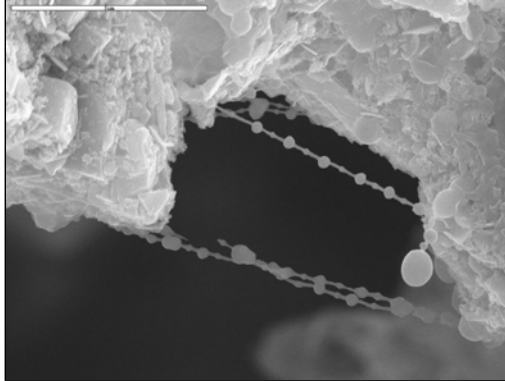
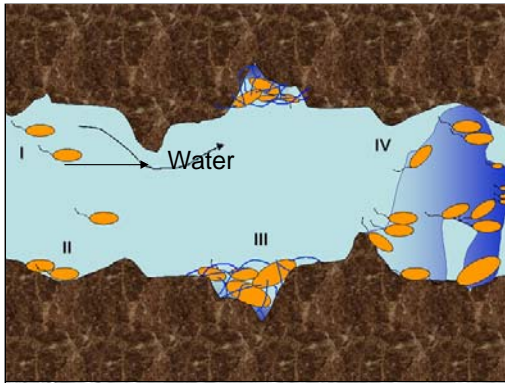
pH of freshwater lens = ~7.6 - 7.2

pH of marine water = ~6.9 - 7.5 ! pH of open ocean is 8.5

NB: In carbonate container, not pH 10

Where is the acid coming from?

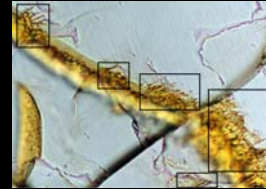
Presenter's Notes: A quick preliminary examination of their claims--applying in nature what is claimed in text books. However, we found something different. BIG problem: acid is not a constant source



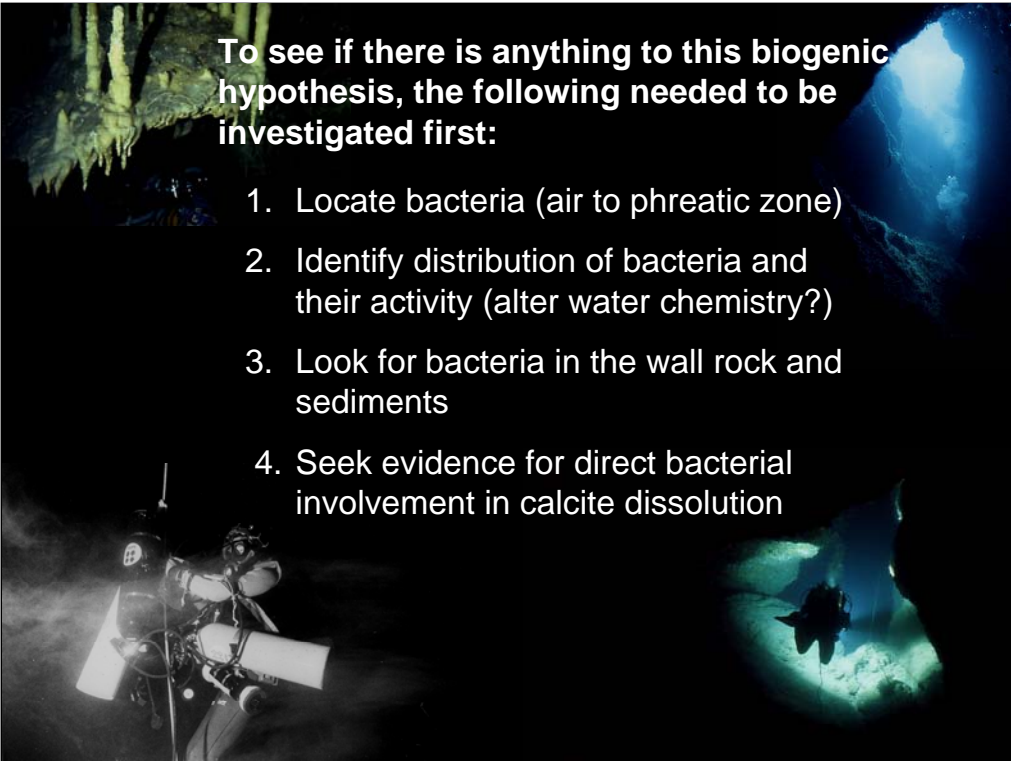
Acids made by bacteria and fungi:

Carbonic acid, oxalic, gluconic, nitric and sulfuric (provide and continues assault on the host rock (surface and interior))

Oxalic acid is 10,000 times stronger than acidic acid....



Presenter's Notes: Hole-making bacteria in your mouth?? Teeth are a mineral. Important: the acid source has to be continuous to make holes, whatever the size. As a result, we have come up with a biogenic hypothesis that may solve, or assist in solving, the problem.



To see if there is anything to this biogenic hypothesis, the following needed to be investigated first:

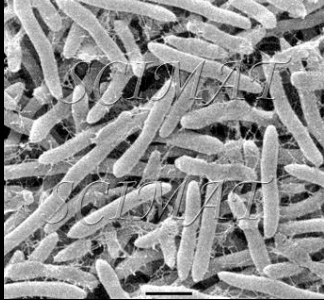
1. Locate bacteria (air to phreatic zone)
2. Identify distribution of bacteria and their activity (alter water chemistry?)
3. Look for bacteria in the wall rock and sediments
4. Seek evidence for direct bacterial involvement in calcite dissolution



..we start with the rain...

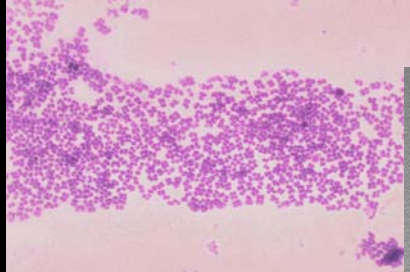
Its Raining Bacteria!

So far, 4 different isolates:

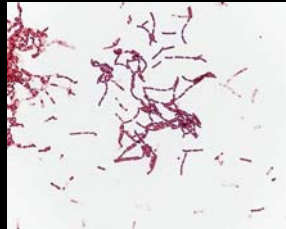


Bacillus megaterium

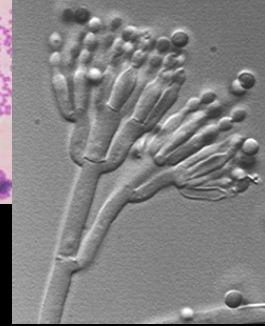
**1000 bacterial cells / ml
of rainwater**



Micrococcus

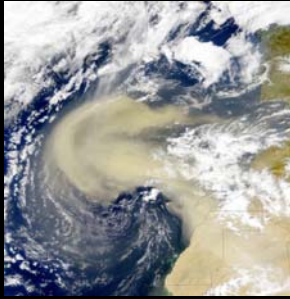


Klebsiella



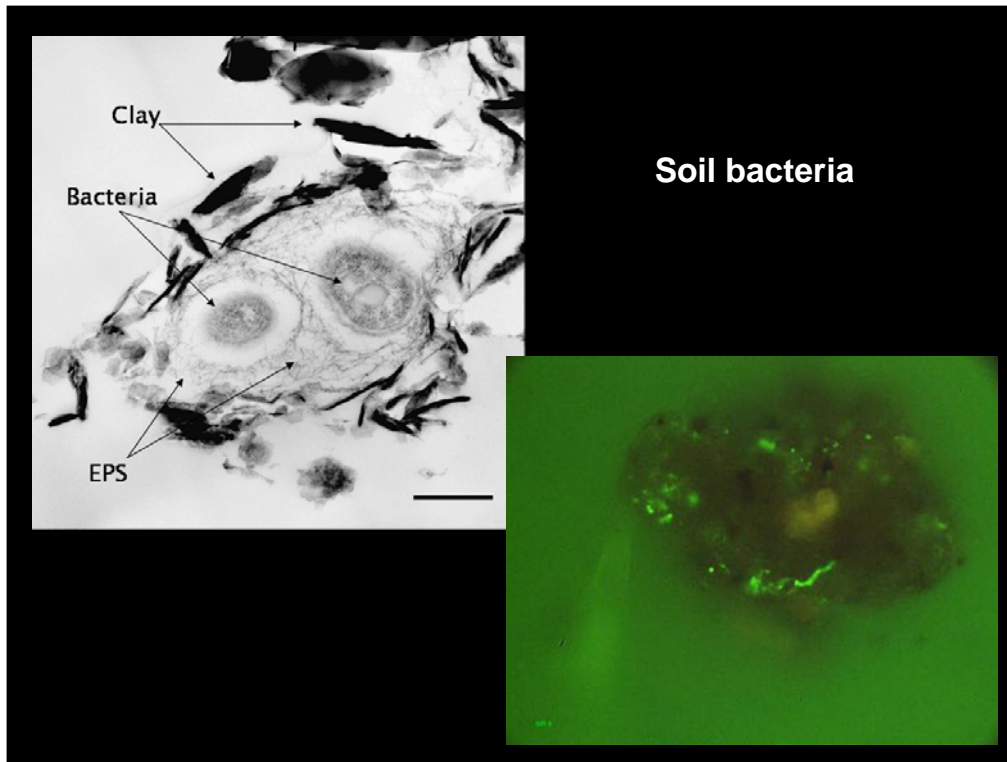
Penicillium

Presenter's Notes: 1 cubic centimeter of dry air from the surface of the Earth to the top of the atmosphere weighs 1.04kg as it relates to CO₂ pp. Why are they (bacteria) in the air?



Dust carries between a million and a billion microbes ($\frac{1}{4}$ tsp).

>300 kinds of microorganisms cultured from air samples collected during Saharan dust storm events. This is approximately 2 to 3 times more than during non-dust events. Of the cultureable bacteria identified thus far, 10% are known animal pathogens, 5% are plant pathogens, and 27% are opportunistic human pathogens (Garrison et al., 2003).



Presenter's Notes: Working our way down from the air towards the phreatic zone. EPS can wedge apart rocks, block subsurface flows; 9 billion bacterial cells per gram of soil on average; 1 billion per ml of ocean water.

Epifluorescent AODC, Acridine Orange Direct count (DAPI)

**Bacteria in rocks in dry
caves?**



CO₂ Flux Meter

CO₂ coming out of the rock is on initial examination 2 to 3 fold higher than what is coming out of the cave floor sediments.

$\delta^{13}\text{C}$ (DIC) values suggest biological source

~52g/m²/day (r² value 0.74)



Presenter's Notes: Unconventional use of a CO₂ flux meter, but recent results are considered to be reasonable when samples are incubated in oxygen free-environments.

New Findings in dry caves!

Cave wall rock incubated in anaerobic environment resulted in the dominant gas production being N_2 , (8.2×10^{-7} to 2.1×10^{-4} μM per g). The next interesting discovery was the production of N_2O . Concentration of this gas was insufficient to measure isotopically but the N_2 signature ranged from -2 to 3 ‰, while the carbon isotope composition of CO_2 ranged from -29 to -6‰

**Before taking you
Into the phreatic
zone, some
definitions.....**

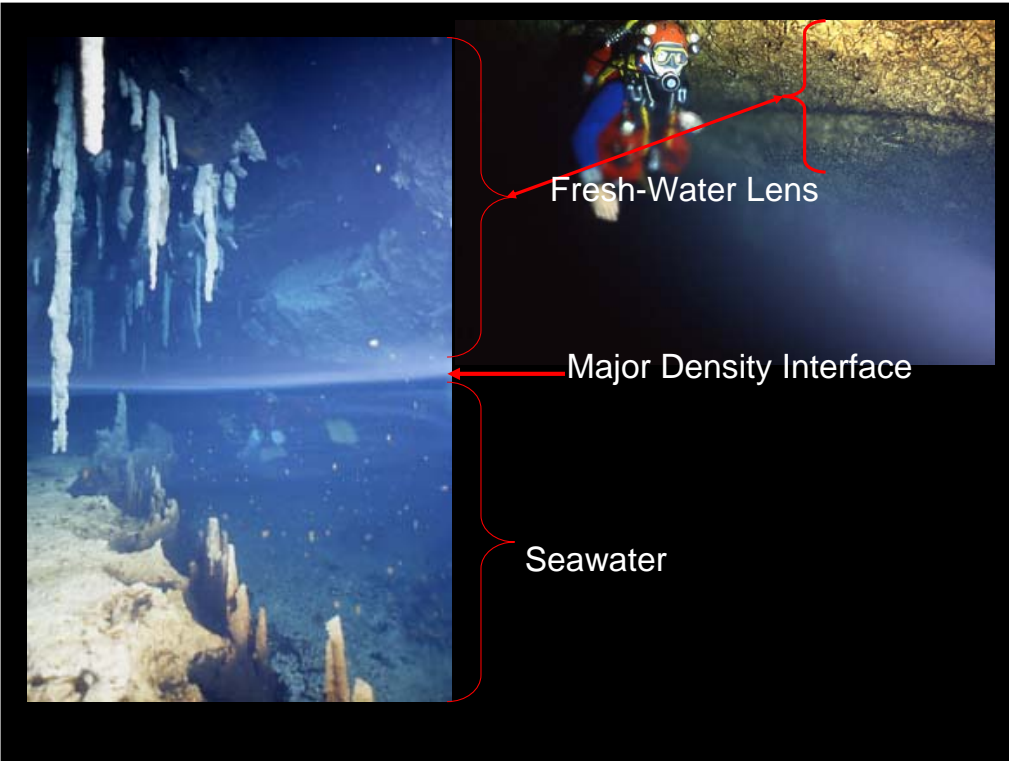
Blue Holes ?

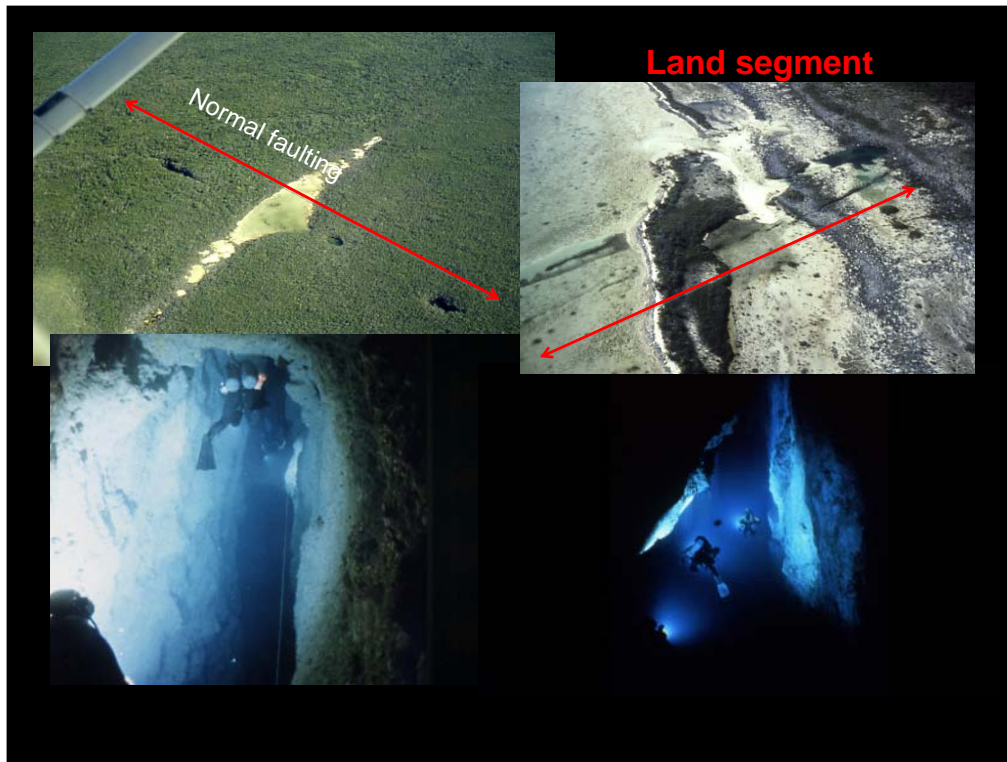


Basic Classification of Caves in the Bahamas

1. Horizontal Caves
2. Fracture-guided Caves
3. Vertical Caves

Presenter's Notes: Sea-level drop, hole is no longer blue--reason for better scientific classification along with new information gained from cave diving. No reason to assume that the caves currently above sea-level are different from those that are currently flooded.





Presenter's Notes: What it looks like from the air and underwater; bank margin failure.

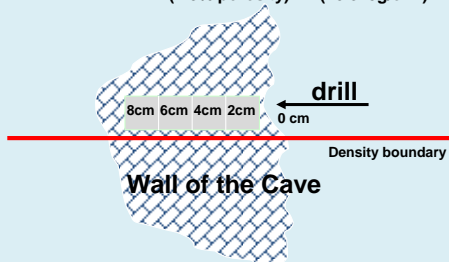
Basic Classification of Caves in the Bahamas

1. Horizontal Caves
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3. **Vertical Caves**

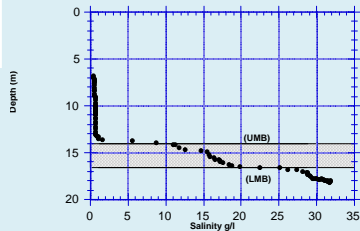
**Bacteria in wall rock,
sediments and
water?.....**

Rock

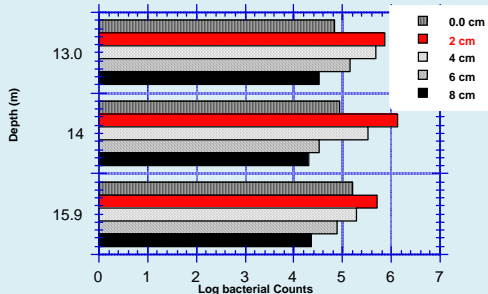
13.0m vertical depth (10-15% porosity) (~0.963g/cm³)
 14.0m (20-30% porosity) (~0.631g/cm³)
 15.9m (1-3% porosity) (~0.815g/cm³)



Wedding Hall, Lucayan Caverns
 Salinity Profile (Hydrolab) 98

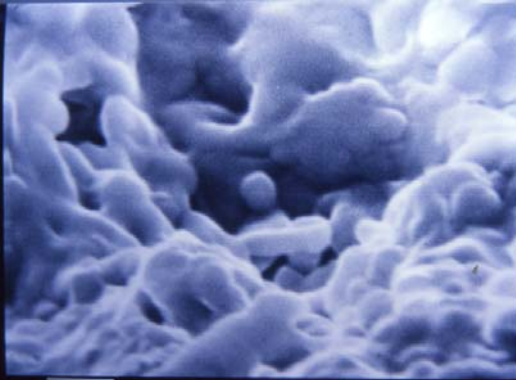


Bacterial Counts within 2.0cm Rock Sections

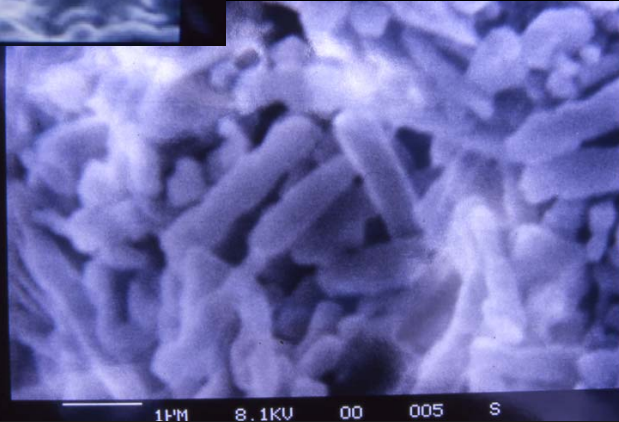


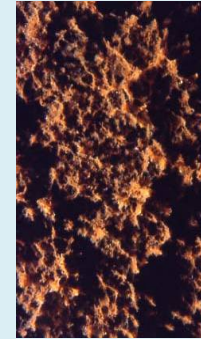
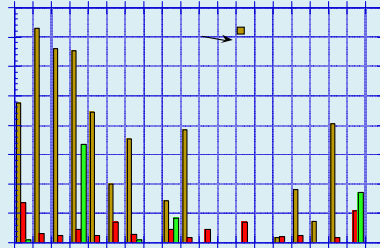
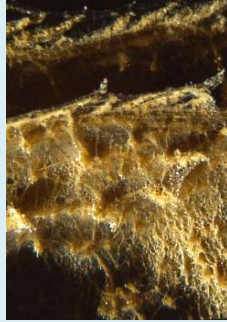
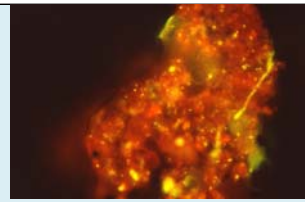
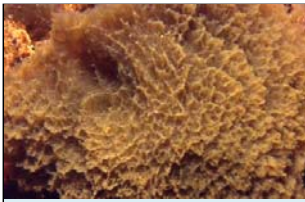
Horizontal Cave: Lucayan Caverns

SEM



Biofilm in a pore of
a **rock**: Note
bacteria beneath the
film





✓ = aliphatic



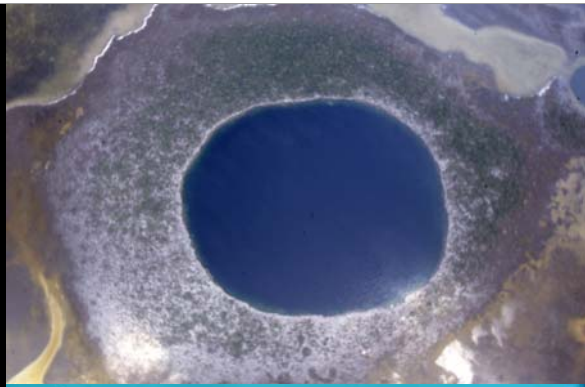
Presenter's Notes: Non-aromatic ring in their structure. Most are flammable (e.g., methane).

Vertical cave:
The Black Hole

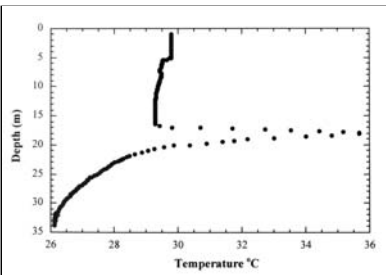


Total mass of purple
bacteria ~ 5 tons dry
weight!

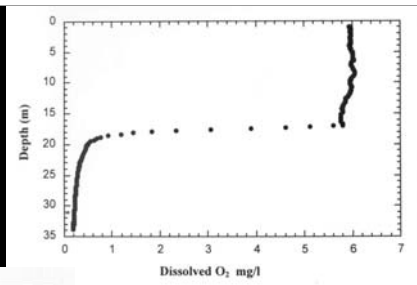
Can these numbers
change the water
chemistry?



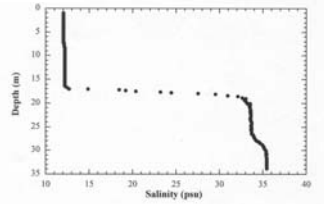
Presenter's Notes: Geochemical changes in the water column.



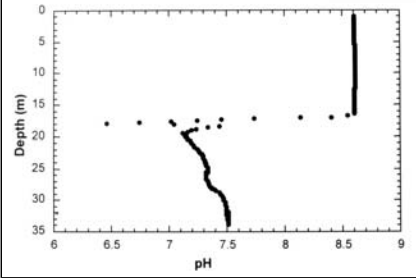
Vertical
Cave:
Black Hole



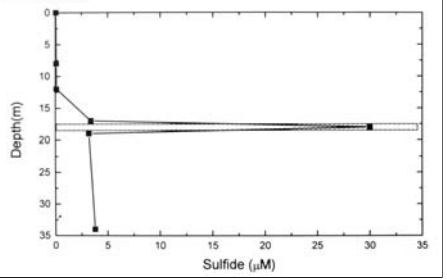
Geochemical Profiles



Water column



Black Hole



**Gas, another way to make
holes?.....**



Berry Islands

Joulters Cay,
North Andros

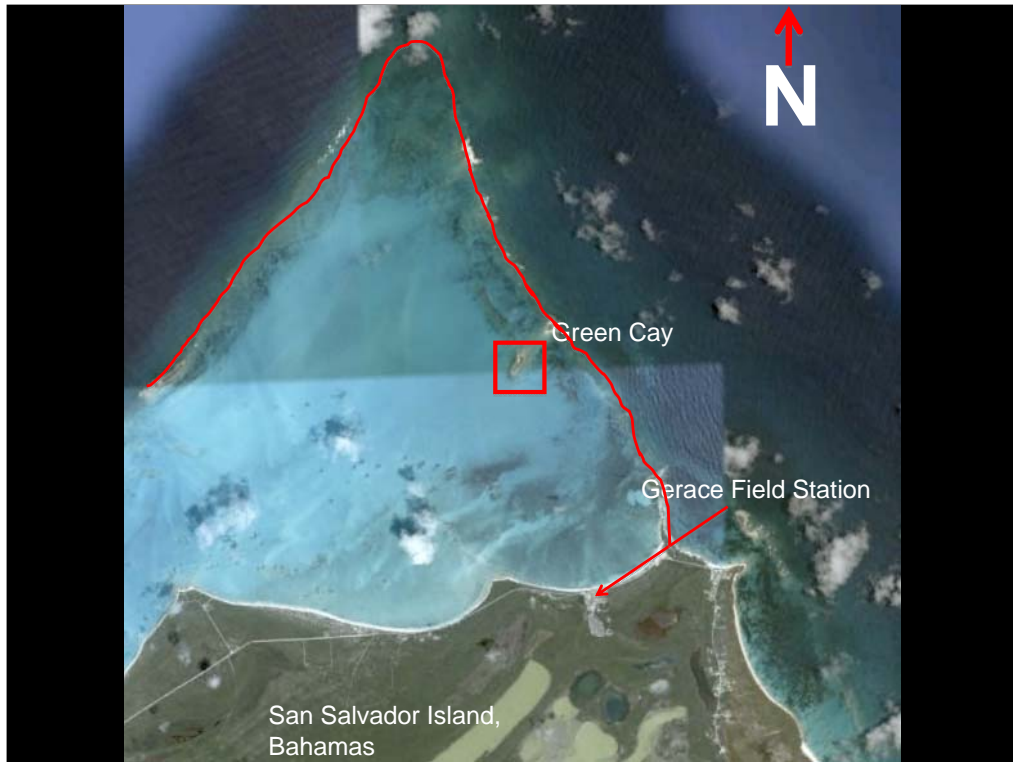
Image U.S. Geological Survey



Presenter's Notes: The sound of popping sand; found items decomposing, and sand was dark purple under the item.

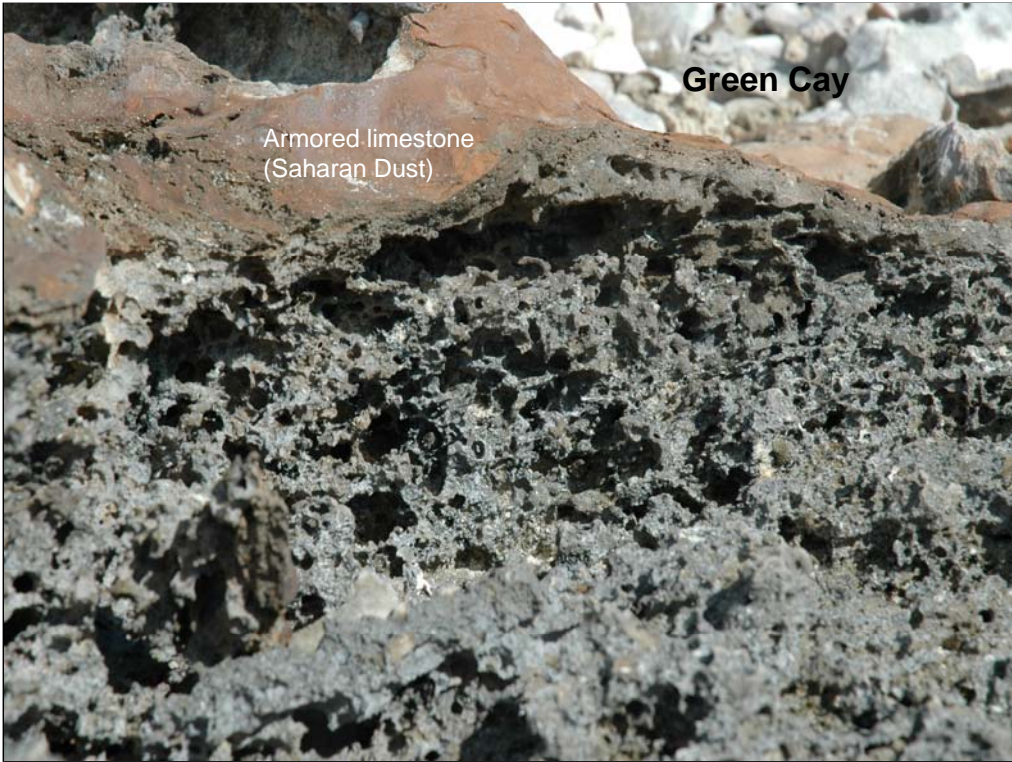


Presenter's Notes: Fermenting? CO₂ signature is pointing in that direction; H₂, NH₃, ethanol, ethylene.....



Green Cay

Armored limestone
(Saharan Dust)



Green Cay (~125,000 yrs) frothy sand?



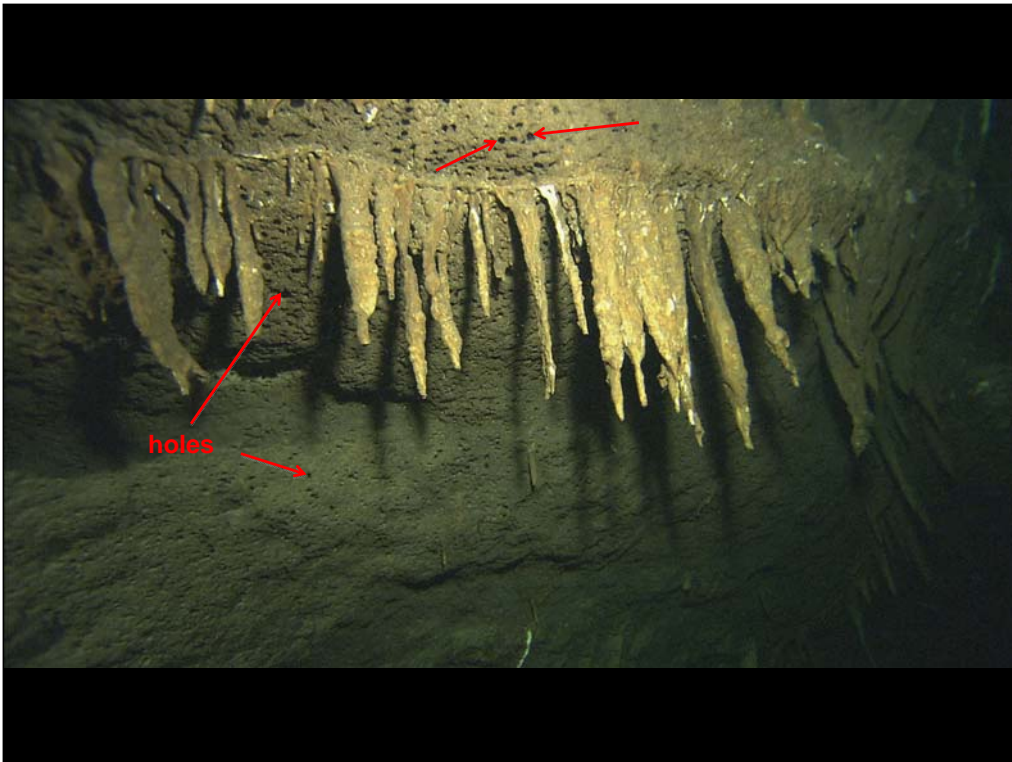
The same?



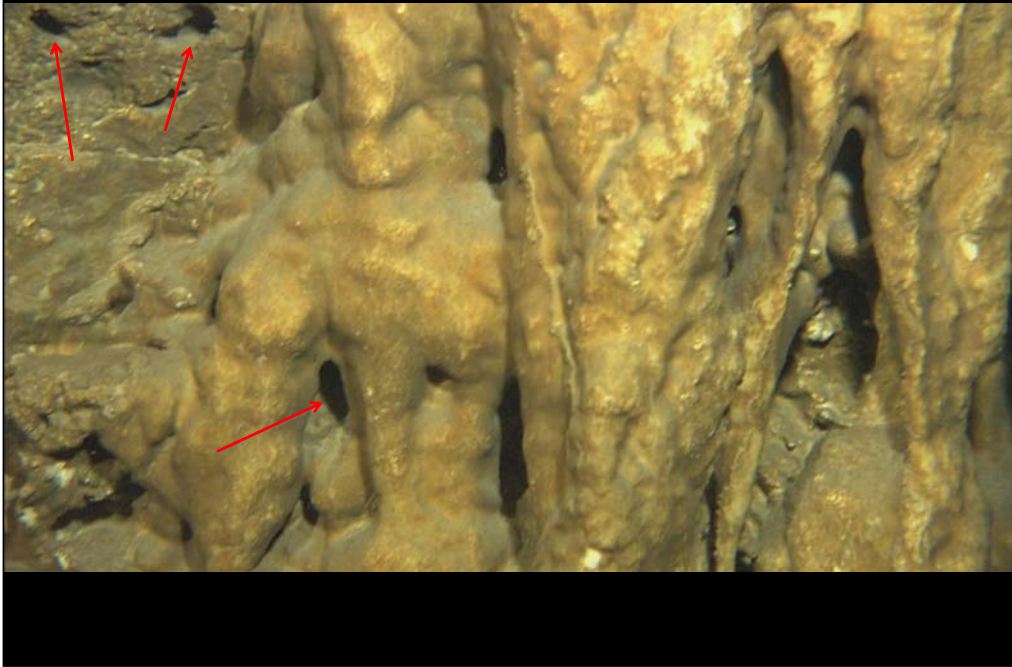
Holocene frothy sand



Presenter's Notes: Go under the mangroves.

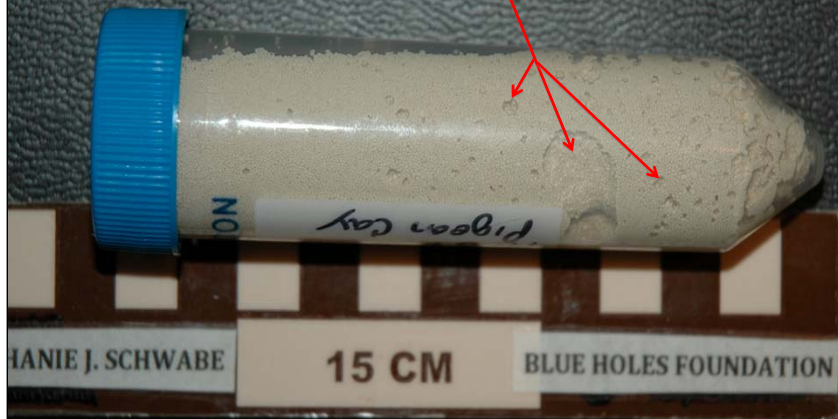


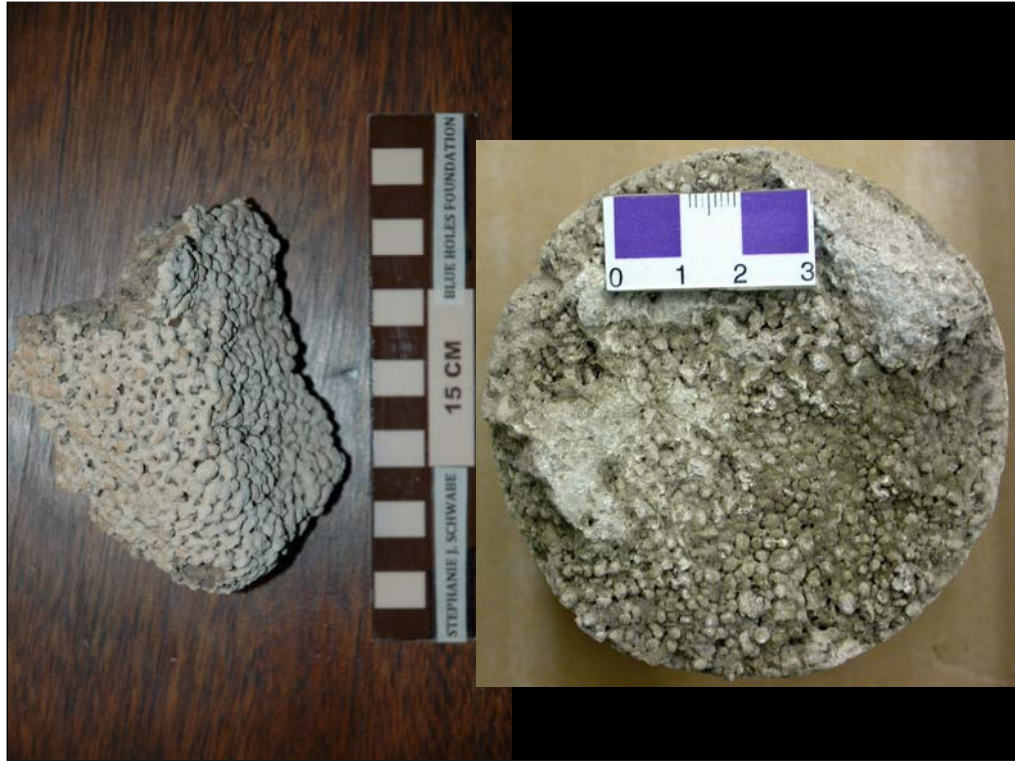
holes



Carbonate sands from Pigeon Cay, Eleuthera Bahamas

Gas bubbles



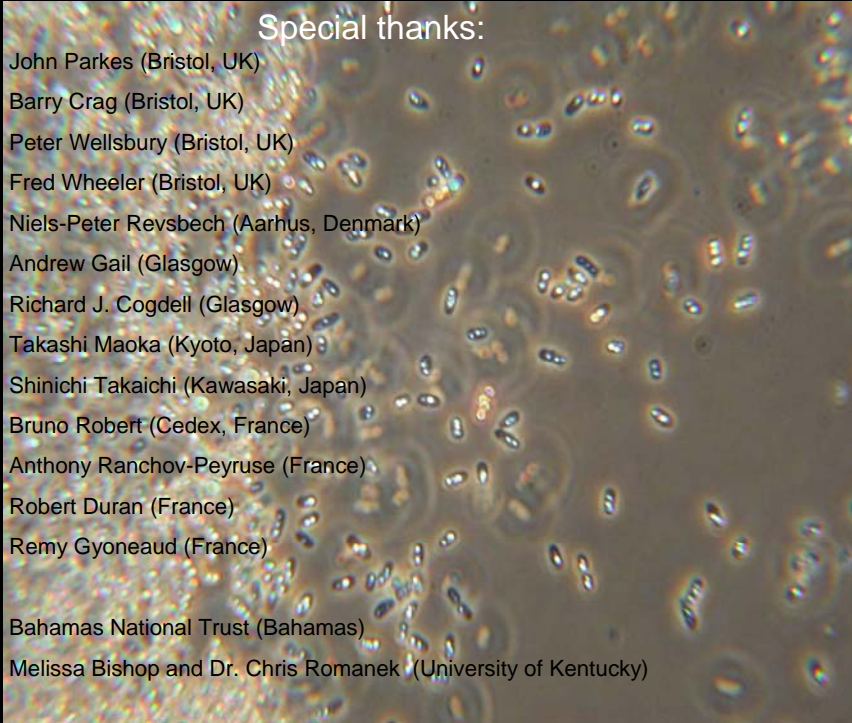






San Salvador Island, Bahamas, Graham's Harbour

Presenter's Notes: Although we consider this research novel, by 1904 scientists had already discovered that interstitial bacteria on the surface and within the rock were responsible for the disintegration of building stones in England. Later in 1933 Pain et al. conducted a very detailed research project that identified the presences of bacteria and the acids they produced were identified. The argument of this process only being a tropical event is null and void. They called it bacterial disease of stones. We call it "the biogenic hypothesis."

A background image showing a dense field of small, oval-shaped cells, likely yeast or bacteria, under a microscope. The cells are scattered across the frame, with some appearing in small clusters. The lighting is somewhat uneven, with brighter areas where the cells are more concentrated.

Special thanks:

John Parkes (Bristol, UK)

Barry Crag (Bristol, UK)

Peter Wellsbury (Bristol, UK)

Fred Wheeler (Bristol, UK)

Niels-Peter Revsbech (Aarhus, Denmark)

Andrew Gail (Glasgow)

Richard J. Cogdell (Glasgow)

Takashi Maoka (Kyoto, Japan)

Shinichi Takaichi (Kawasaki, Japan)

Bruno Robert (Cedex, France)

Anthony Ranchov-Peyruse (France)

Robert Duran (France)

Remy Gyoneaud (France)

Bahamas National Trust (Bahamas)

Melissa Bishop and Dr. Chris Romanek (University of Kentucky)