#### Tracer Gas Diffusion in the Eagle Ford Shale, Austin Chalk, and Adjunct Vertical Formations in Southwestern Texas\*

Qiming Wang<sup>1\*</sup>, Qinhong Hu<sup>1</sup>, and Xiang Lin<sup>2</sup>

Search and Discovery Article #42568 (2021)\*\*
Posted February 15, 2021

#### **Abstract**

As one of the two critical transport mechanisms in shale gas reservoirs, gas diffusion can be quantified by the diffusion coefficient (m²/s) within the shale matrix. To understand the diffusion behavior in rock matrices, 1-D short-duration (within 24 hours) tracer gas diffusion chamber tests at a room temperature were conducted on the major reservoirs (Eagle Ford B calcareous Shale and Austin Chalk), and adjunct vertical formations (Atoc Chalk, Buda Limestone, Eagle Ford A dolomitic ash bed, and Salmon Peak Limestone) in the Southerwetern Texas area. Associated with X-ray diffraction, thin section, and mercury intrusion porosimetry, the mineral composition, pore structure (both geometry and connectivity) were taken into the discussion of influencing factors. The results of gas diffusion tests show that the diffusion coefficients among these rocks with different lithologies vary in the magnitude of 10<sup>-8</sup> to 10<sup>-7</sup> m²/s and is influenced by pore structure especially pore connectivity.

#### References

Currie, J.A, 1960, Gaseous diffusion in porous media. Part 2. Dry granular materials. Br. J. Appl. Phys. 11:318–324.

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<sup>\*</sup>Adapted from oral presentation accepted for the 2020 AAPG Annual Convention and Exhibition online meeting, September 29 – October 1, 2020.

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# Tracer Gas Diffusion in the Austin Chalk, Eagle Ford Shale and Adjunct Vertical Formations in Southwestern Texas

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The University of Texas at Arlington



Boquillas (Eagle Ford) outcrop, Del Rio, TX

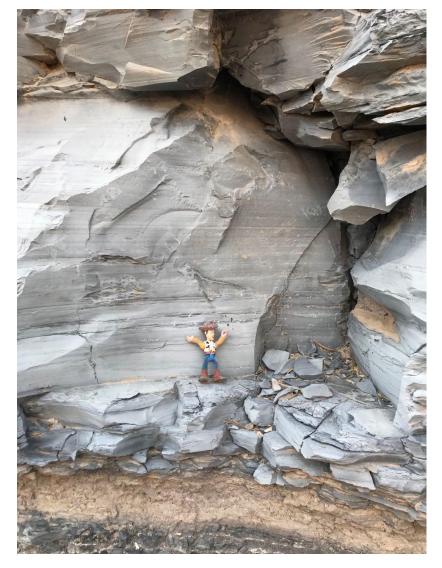




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#### **Outline**

- Diffusion in natural rocks
- Sample properties
- Tracer gas diffusion method
- Summary



Eagle Ford B Calcareous Shale



#### Diffusion in natural rocks



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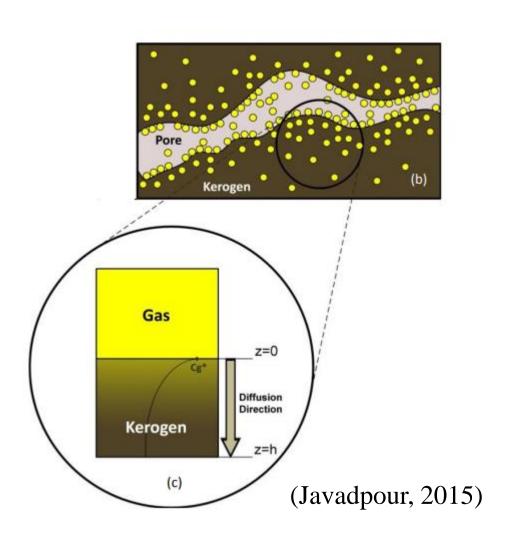
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#### Why diffusion is important?

- Occur at gas or liquid phase
- Rate-limiting or dominate process
   of fluid flow and mass transport in lowpermeability geological media

#### Distinguishing features

- Random particle walk
- Driven by concentration gradient, influenced by temperature





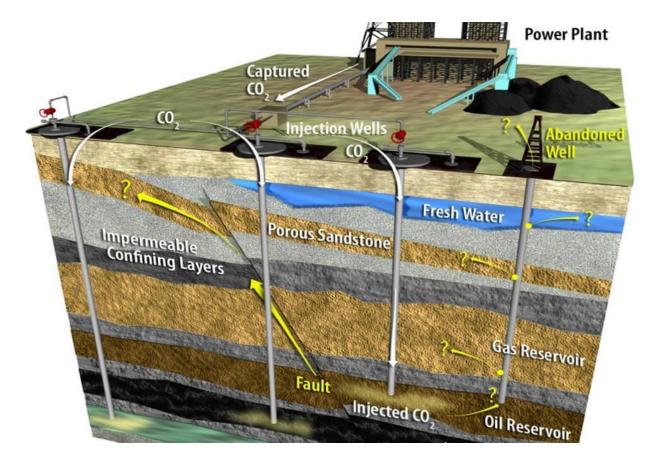
#### Diffusion in natural rocks



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#### Applications

- Oil and gas recovery
- CO<sub>2</sub> sequestration
- Contaminant remediation
- Geologic disposal of radioactive waste



(GoldSim)



# Sample properties





Series	Stage	Formation	
Upper Cretaceous	Santonian	Austin Chalk	Outcrop in Midlothian, Ellis County, T
	Coniacian	Atco Chalk	
	Turonian	E D Eagle Ford Fm. Boquillas Fm.	Langtry  TEXAS  USA
	Cenomanian	B A Buda Limestone Del Rio Formation	Atco Chaik  Eagle Ford B Calcareous Shale  Eagle Ford A Dolomitic Ash Bed  Buda Limestone  Salmon Peak Limestone  City/Town  Highway  River/Lake  Borderline
Lower	Albian	Salmon Peak Limestone	Texas MEXICO Del Rio

Del Rio, Val Verde County, TX

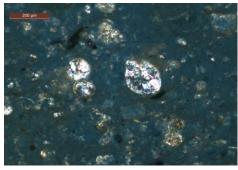


# Sample properties

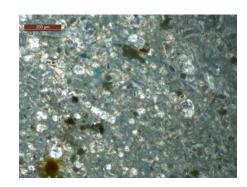


# TEXAS ARLINGTON

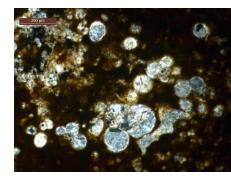
Sample ID				Mineral o	compositio	n (wt.%)				
Stample 12	Quartz	Calcite	Ankerite	Kutnohorite	Pyrite	Magnetite	Goethite	Kaolinite	Illite	_
Austin Chalk	1.2	96.3		0.5					2.0	
Atco Chalk	1.0	99.0								
Eagle Ford B Calcareous Shale	15.5	79.6	1.2		0.8			1.3	1.6	
Eagle Ford A Dolomitic Ash Bed	9.8	0.7	44.9	36.5		0.2	2.0	2.8	3.2	
Buda Limestone	1.3	98.7								
Salmon Peak Limestone	0.2	99.8								_



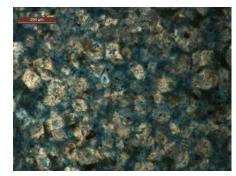
Austin Chalk



Atco Chalk



Eagle Ford B
Calcareous Shale



Eagle Ford A
Dolomitic Ash Bed



**Buda Limestone** 



Salmon Peak Limestone



### Sample properties

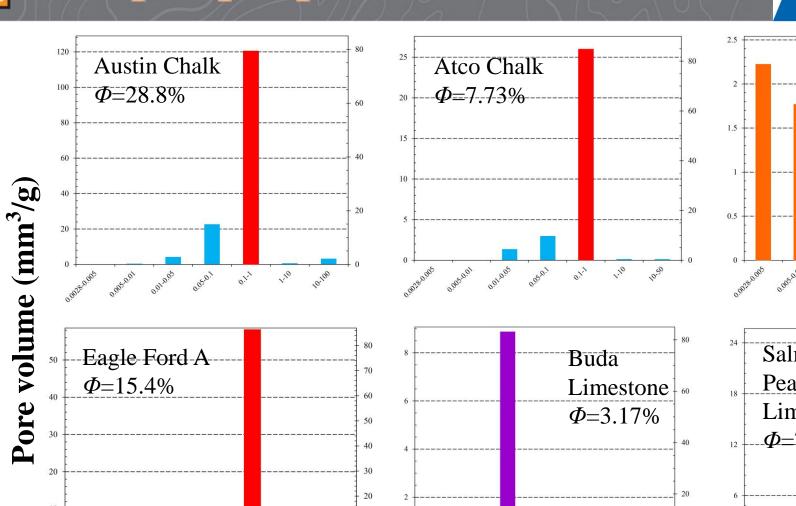


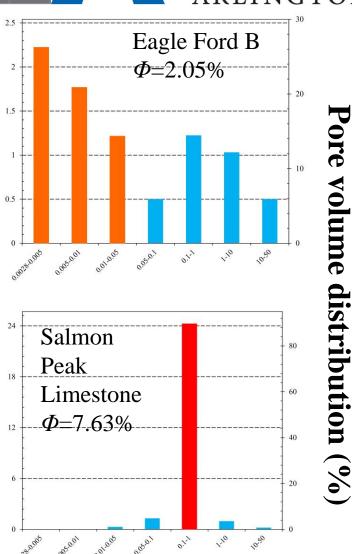
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This part is

Mercury intrusion porosimetry





Pore-throat diameter (µm)





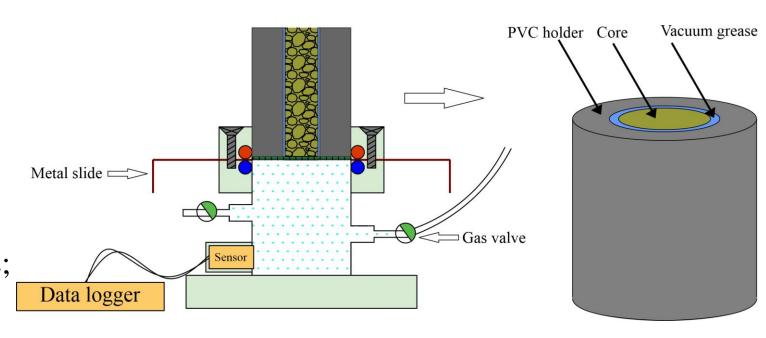
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#### Currie method:

- First reported by Currie (1960).
- Commonly used in soil science
- Tracer gas: O<sub>2</sub>

#### Advantages:

- Various sample shapes (irregulars; regulars: cylindrical, cubic, and granular)
- Applicable to various initial sample conditions (oven-dry, airdry, partially saturated, fully saturated)





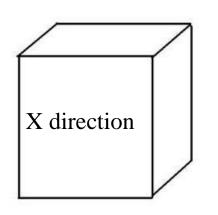


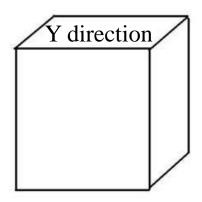
PVC holder Core

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Vacuum grease





Diffusion coefficients analyzed in two directions perpendicular to each other

Sample and holder was sealed by vacuum grease to minimize leakage







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#### Fick's First law-based diffusion equation

$$C_r = \frac{C_t - C_s}{C_0 - C_s} = \sum_{n=1}^{\infty} \frac{2h \exp\left(-\frac{D_p \alpha_n^2 t}{\emptyset}\right)}{L(\alpha_n^2 + h^2) + h}$$

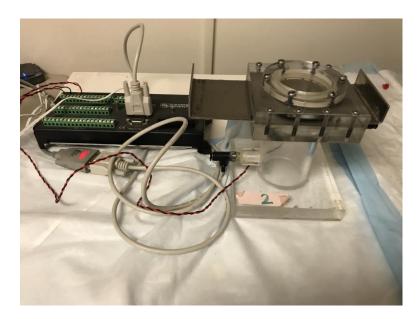
(Rolston and Moldrup, 2002)

#### At ln-ln scale,

$$lnC_r = -\frac{D_p \alpha_1^2 t}{\emptyset} ln \left( \frac{2h}{L(\alpha_1^2 + h^2) + h} \right)$$

In  $C_r$  is a linear function to t with a slope of  $-\frac{D_p \alpha_1^2}{\emptyset}$ 

Diffusion coefficient  $D_p$  could be determined when  $\alpha$  and  $\emptyset$  is known.



- C<sub>r</sub>: tracer gas concentration
- Ct: tracer gas concentration in the chamber when t=t
- C0: tracer gas concentration in the chamber when t=0
- Cs: tracer gas (O<sub>2</sub>) concentration in atmosphere
- h=Ø/a
- A: the length or volume of the diffusion chamber or volume of chamber per area of the sample
- Dp: diffusion coefficient of sample to tracer gas
- α<sub>n</sub>: the positive roots of α<sub>n</sub>tan(α<sub>n</sub> L)=h, with n=1,2,.... When t >0, the terms for n ≥ 2 are negligible due to the very small influence on the result when compared to n=1

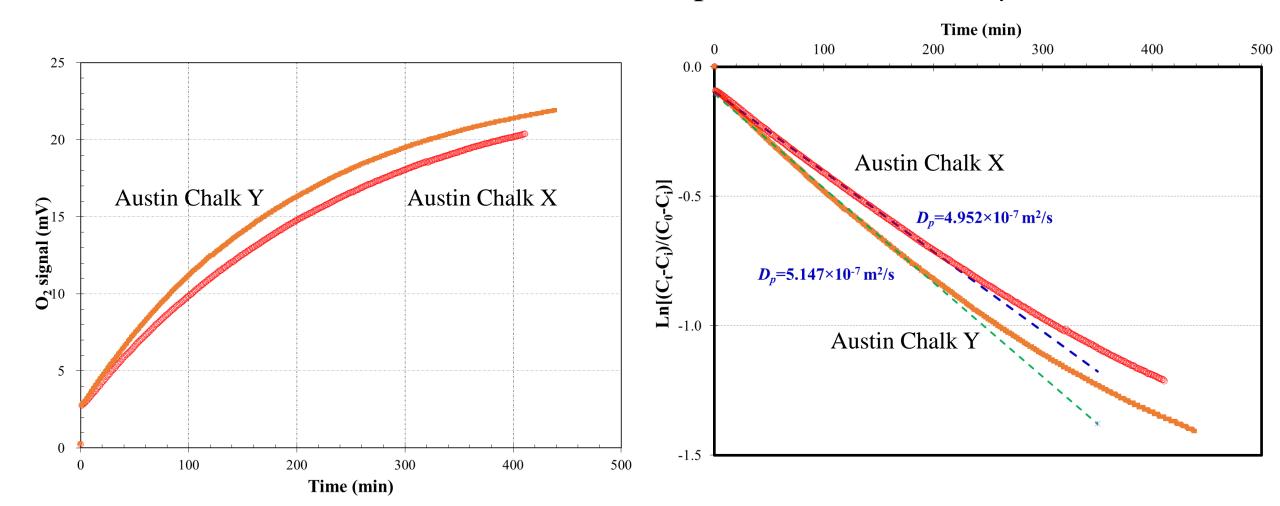




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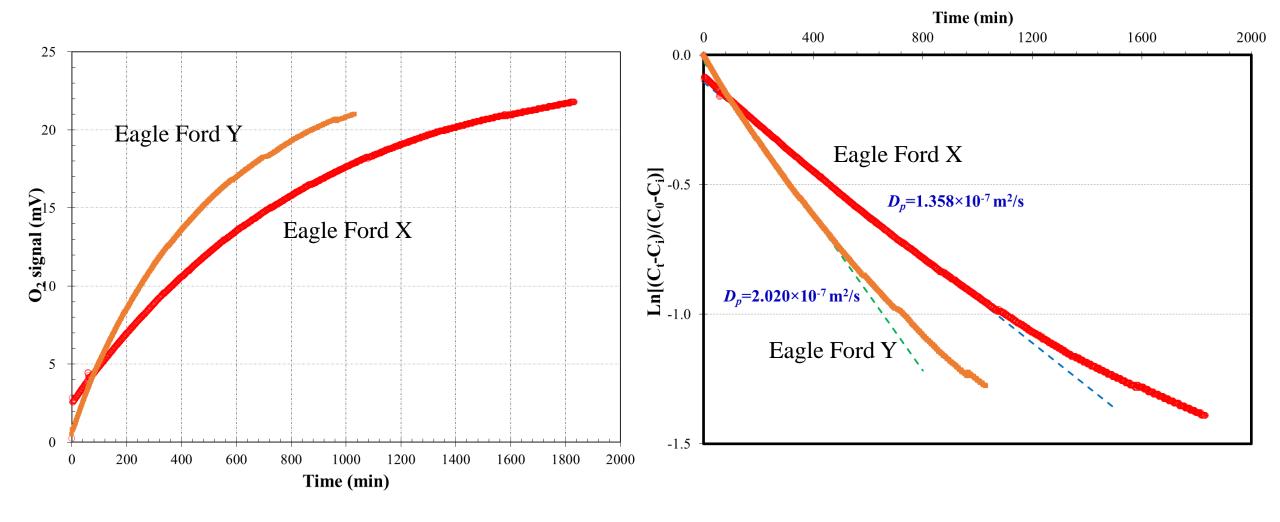
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Austin Chalk  $\Phi$ =28.8% Dominate pore diameter: 0.1-1µm



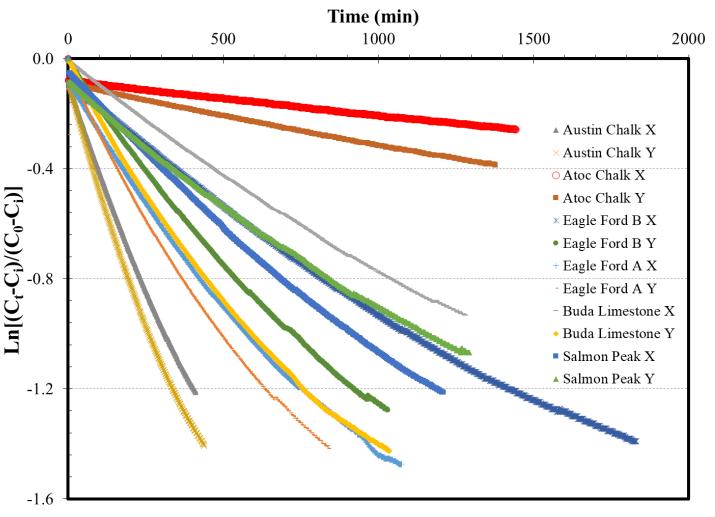


#### Eagle Ford B Calcareous Shale $\Phi$ =2.05%





#### The results show directional heterogeneity



G 1 ID	Direction X	DirectionY		
Sample ID	$D_p  (\mathrm{m}^2/\mathrm{s})$	$D_p  (\mathrm{m}^2/\mathrm{s})$		
Austin Chalk	4.952E-07	5.147E-07		
Atco Chalk	4.895E-08	3.427E-08		
Eagle Ford A Dolomatic Ash Bed	2.417E-07	3.110E-07		
Eagle Ford B Calcareous Shale	1.345E-07	2.020E-07		
Buda Limestone	1.217E-07	3.012E-07		
Salmon Peak Limestone	1.968E-07	1.137E-07		



# Issues in determining $D_p$

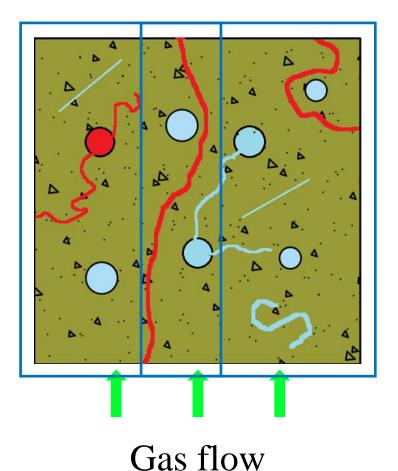
#### At ln-ln scale,

$$lnC_r = -\frac{D_p \alpha_1^2 t}{\emptyset} ln \left( \frac{2h}{L(\alpha_1^2 + h^2) + h} \right)$$

What is the porosity used in the D<sub>p</sub> calculation?

- For monolithic samples: with different pore connectivity
- For granular samples: with interand intra-granular pore space

#### Pores are not 100% interconnected in rock matrix



- Porosity Ø used is the porosity of the whole monolithic rock
- Porosity Ø should be using is the fluid flow porosity in a specific direction

#### Pore connectivity

- Well connected materials: soils, granular rock samples, loose sandstone, and porous carbonate rocks
- Poorly connected materials: tight sandstone, tight carbonate rocks, crystalline rocks, shales, and evaporites

Well connected:

Poorly connected:

$$lnC_r = -\frac{D_p \alpha_1^2 t}{\emptyset} ln \left( \frac{2h}{L(\alpha_1^2 + h^2) + h} \right) \quad \begin{matrix} \bigcirc p_{fluid flow} \approx \bigcirc p_{whole sample} \\ D_{p (true)} \approx D_{p (calculated)} \end{matrix} \quad \begin{matrix} \bigcirc p_{fluid flow} < \bigcirc p_{whole sample} \\ D_{p (true)} \approx D_{p (calculated)} \end{matrix}$$

$$ot\!\!\!/ p_{fluid\,flow} pprox 
ot\!\!\!/ p_{whole\,sample}$$

$$D_{p \, (true)} \! pprox \! D_{p (calculated)}$$

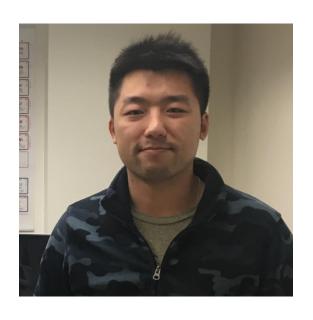
$$D_{p \, (true)} < D_{p (calculated)}$$

- Tracer gas diffusion method to determine the diffusion coefficient is applicable to a wide range of rock lithologies, as demonstrated in a vertical profile in Texas.
- Porous Austin Chalk has a porosity of 28.8% and average diffusion coefficient of 5.050×10<sup>-7</sup> m<sup>2</sup>/s.
- Tight Eagle Ford B Calcareous Shale has a porosity of 2.05% and average diffusion coefficient of 1.683×10<sup>-7</sup> m<sup>2</sup>/s.
- Diffusion coefficient will be overestimated if an incorrect porosity, which is related to pore connectivity, is used in the calculation.



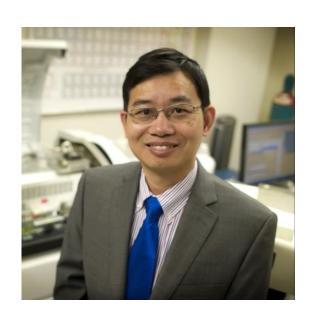
## Acknowledgements





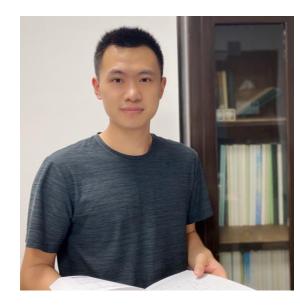
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- R.E. McAdams Memorial Grant from the AAPG Foundation's Grants-in-Aid Program
- The National Natural Science Foundation of China (No. 41672251).

