#### **Organic Shale NMR and What It Means for Logs\***

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#### Conclusions

- Brine is mainly in water-wet pores.
- Methane is mainly in gas-wet pores.
- Surface relaxivity dominates both signals.
- No evidence exists that methane will occupy pore space that holds water.
- NMR gas porosity is equal at least to free-gas porosity.
- Water saturation is 30 to 45 %.
- Diffusion will have only minor effect on primary part of methane T<sub>2</sub> spectrum.
- NMR T<sub>1</sub> log spectrum will not show methane at bulk relaxation time.

#### References

Sigal, R., and E. Odusina, 2011, Laboratory NMR Measurements on Methane Saturated Barnett Shale Samples: MPGE, University of Oklahoma, Petrophysics.

Straley, C., 1997, An Experimental Investigation of Methane in Rock Materials: SPWLA 38th Annual Logging Symposium, Paper AA, 14 p.

Winkler, M., J.J. Freeman, E. Quint, and M. Caputi, 2006, Evaluating Tight Gas Reservoirs with NMR – The Perception, The Reality and How to Make It Work: SPWLA 47th Annual Logging Symposium, Paper BB, 9 p.

# Organic Shale NMR and What it Means for Logs

Richard Sigal Elijah Odusina University of Oklahoma AAPG Workshop 2010

### Good and Bad news

- Good News
  - The NMR log porosity contains a contribution from almost all to all the free gas in a shale gas reservoir.
  - The gas distribution may be able to be used to estimate adsorbed gas.
- Bad News
  - Gas signal probably cannot be extracted from standard NMR log
  - Using standard processing to identify gas signal will give incorrect answer.

# **Ordinary Reservoirs**

- Pore walls not gas wetting
- Measurement made in invaded zone
- Relatively large pores
- Surface relaxation dominated by paramagnetic minerals
- Implies for T<sub>2</sub> measurement
  - High-wetting fluid saturation
  - Gas in centers of pores
  - Diffusion coefficient is approximately bulk value
  - $1/T_{gas} = 1/T_{bulk} + 1/T_{diffusion}$
  - Logging tools have magnetic field gradient so that  $\rm T_2$  gas is about equal to  $\rm T_{diffusion}.$

## If Previous Slide Incorrect

Low water saturation

Gas can have surface relaxation mode Straley (SPWLA 38, 1997) Winkler et al. (SPWLA 47, 2006) Diffusion may be less important Small pores

Diffusion coefficient not bulk value

Organic pores

Small

Pore walls hydrophobic

Probably no water invasion

Surface relaxation component

Relaxation may not be dominated by paramagnetic minerals

Relaxation could be temperature-dependent

Adsorbed gas

## **Organic Shales**

- Some to most porosity in organic materials
- Small pores
- Possibly very limited invasion from waterbased mud
  - Low permeability
  - Organic pores hydrophobic

## **Organic Pores**

- Organic Porosity
  - SEM images show high porosity in some organic material
  - Petrophysical calculations give roughly 20 to 50% organic porosity.
- Pore Size
  - SEM 1-100 nm range very roughly
  - Surface to volume estimate
    - Barnett
    - Langmuir isotherm
    - Gas volume
    - Effective pore size on order of 2-20 nm.

# Diffusion

- Zones of organic material from SEM on order of microns.
- Methane diffusion distance in 1 ms
  - Methane bulk diffusion constant
  - 4000 psi, 355°C
  - 10,000 nm =10  $\mu$

#### Lab Measurements

Barnett samples Room temperature 5000 psi confining pressure 4000 psi pore pressure

#### Preserved State No Methane T<sub>2</sub> Response No External Magnetic Field Gradient



#### Methane T<sub>2</sub> Response Barnett Preserved State No External Magnetic Field Gradient



| Sample #          | NMR                 | NMR                   | Не              | Water          | State |
|-------------------|---------------------|-----------------------|-----------------|----------------|-------|
|                   | Gas Porosity<br>(%) | Water Porosity<br>(%) | Porosity<br>(%) | Saturation (%) |       |
| Well 1<br>6366.78 | 4.06                | 3.52                  | 7.16            | 46.4           | Р     |
| Well 1<br>6366.58 | 4.10                | 2.86                  | 6.26            | 41.1           | Р     |
| Well 1<br>6454.95 | 4.50                | 1.74                  | 6.61            | 27.9           | Р     |
| Well 2 7529.6     | 5.98                | 2.57                  | 6.24            | 26.9           | Р     |
| Well 2 7433       | 4.95                | 2.92                  | 8.45            | 37.1           | Р     |
| Well 3 7259       | 3.47                | 1.02                  | 7               |                | U     |
| Well 3 7508       | 3.92                | 3.18                  | 8               |                | U     |
| Well 3 7591       | 4.72                | 1.67                  | 9               |                | U     |
| Well 3 7621       | 3.61                | 1.56                  | 7               |                | U     |
| Well 3 7694       | 2.7                 | 1.56                  | 6               |                | U     |

#### Observations

- Dominate brine response less than 1 ms.
- Usually brine has secondary mode
- At room temperature dominate gas response in 1-20 ms range
- About 10% of gas response comes in at bulk decay rate for methane. This is gas in stress release cracks.
- Unpreserved samples have no more gas porosity than preserved samples.

## Conclusions

- Brine is mainly in water-wet pores
- Methane mainly in gas-wet pores
- Surface relaxivity dominates both signals
- No evidence that methane will occupy pore space that holds water
- NMR gas porosity at least equal to free-gas porosity
- Water saturation 30 to 45 %
- Diffusion will have only minor effect on primary part of methane T<sub>2</sub> spectrum
- NMR T<sub>1</sub> log spectrum will not show methane at bulk relaxation time.

## **Open Questions**

- Amount of adsorbed gas in NMR gas porosity.
- Restricted diffusivity value.
- Best way to separate gas and brine in logging signal.
- The effect of temperature on gas relaxivity.
- How typical is Barnett NMR?

#### Selected Reference

Laboratory NMR Measurements on Methane Saturated Barnett Shale Samples, Richard Sigal and Elijah Odusina, MPGE, University of Oklahoma, Petrophysics 2011