PSIn-Situ, Micron-Scale Analyses of Carbonate δ13C by SIMS: A Method for Identifying Carbonate Cements that Form in Response to CO₂ Injection at the Illinois Basin Decatur Project Site*

M. G. Sliwinski¹, K. Kitajima¹, R. Kozdon^{1,2}, A. Denny¹, M. J. Spicuzza¹, and J. W. Valley¹

Search and Discovery Article #51325 (2016)**
Posted November 7, 2016

Abstract

This work demonstrates the utility of in situ carbon and oxygen isotope microanalysis (δ^{13} C and δ^{18} O) by secondary ion mass spectrometry (SIMS) in carbon sequestration research. A desirable long-term consequence of CO₂-injection into underground rock formations at prospective storage sites (such as deep saline sandstone aquifers capped by impermeable strata) is the precipitation of carbonate mineral cements, the isotopic fingerprinting of which is a central theme here. More specifically, we focus on the unique advantage of the SIMS technique, which lies in the capability of analyzing very small sample volumes that are otherwise inaccessible to sampling techniques in conventional isotope ratio mass spectrometry (IRMS). For example, single carbonate crystallites as small as 3-10 μ m across can be readily analyzed by SIMS with sub per-mil accuracy and precision. Importantly, the ability to perform micrometer-scale measurements in situ from either thin sections or 1-inch (25 mm) diameter polished core plugs preserves the petrographic context of measured carbonate δ^{18} O and δ^{13} C values.

An important aspect of analytical methods-advancement is the continued development of reference materials that will allow for correcting SIMS-specific sample matrix effects in a wider range of Ca-Mg-Fe-carbonate compositions. At present, routine analyses can be performed on the end-members calcite (and aragonite), siderite and magnesite, as well as on the spectrum of compositions that fall along the dolomite-to-ankerite solid solution series (Śliwiński, et al., 2016a, 2016b).

At the Illinois Basin Decatur Project (IBDP) demonstration site (Leetaru et al., 2009; Leetaru and Freiburg, 2014), it is estimated that at least 10-20% of the introduced CO_2 may be consumed, and thus permanently stored, via the precipitation of Ca-Mg-Fe-carbonates (e.g. Liu et al., 2011; Carroll et al., 2013). We provide here a preliminary characterization of the pre-injection mineralogy and isotopic fingerprints ($\delta^{13}C$ and $\delta^{18}O$) of carbonate cements in the Mt. Simon Sandstone reservoir and the overlying shaly caprock (the Eau Claire Formation). By drawing upon published data on ambient reservoir conditions and the C-isotope composition of the injected CO_2 , we also make simple predictions about the possible $\delta^{13}C$ signatures of different carbonate cement types that may form in response to imposed CO_2 storage.

^{*}Adapted from poster presentation given at AAPG Annual Convention & Exhibition, Calgary, Alberta, Canada, June 19-22, 2016

^{**}Datapages © 2016 Serial rights given by author. For all other rights contact author directly.

¹WiscSIMS, Department of Geoscience, University of Wisconsin, Madison, Wisconsin (msliwinski@wisc.edu)

²Lamont-Doherty Earth Observatory, Columbia University, Palisades, New York

Carbonate cements are generally abundant (although somewhat heterogeneously distributed) within the diverse suite of lithofacies that comprise the Eau Claire (Neufelder et al., 2012; Śliwiński et al., 2016). Cement mineralogy is predominantly dolomite-ankerite, calcite and subordinate siderite to Mg-siderite. Volumetrically significant occurrences also locally occupy the pore space of the upper Mt. Simon near contact with the overlying caprock (Hoholick et al., 1984; Fishman, 1997; Bowen et al., 2011; Denny et al., 2015). Limited analyses indicate that carbonate minerals (calcite, dolomite) are also present in mudstone/shale interbeds at depth in the general vicinity of the CO₂ injection zone within the lower Mt. Simon (e.g. see Table 8.3 in Finley, 2005). These clay mineral-rich interbeds are considered to be the first reactive environments that will be encountered by the emplaced CO₂ plume as it buoyantly rises through the reservoir, and are estimated to have the same capacity for sequestrating carbon by mineral-trapping reactions as the basal Eau Claire shale (Finley, 2005). Characterization could include isotopic fingerprinting studies of carbonates from these intervals in core material recovered from: (1) wells drilled at the IBDP site prior to CO₂-injection (2011-2014), and (2) future coring efforts during the now on-going, post-injection monitoring phase of the project.

Comparative studies of pre- and post-injection samples could aid in ground-truthing and/or refining reactive flow and transport models that attempt to: (1) predict the rate of mineral-trapping reactions, (2) to delimit the likely spatial distribution of reaction products (carbonate cements), and (3) to estimate the fraction of the CO₂ introduced into the reservoir/caprock system that will likely become permanently immobilized in mineral form (Liu et al., 2011). Whereas the CO₂ injection zone is situated within the lower Mt. Simon Sandstone at the IBDP site, injection was also planned into the upper portion of this reservoir unit at the nearby site of the FutureGen 2.0 project (now defunded; e.g. Bonneville et al., 2013).

We identify some areas of current research interest where in situ isotope microanalysis by SIMS could provide unique insights to investigations concerned with understanding how CO₂ will interact with geological reservoirs and caprocks under engineered storage conditions. It is likely that studies of natural CO₂ accumulations - regarded as analogues to engineered CO₂ storage - could find the micrometer-scale spatial resolution afforded by the SIMS technique particularly advantageous.

Research efforts directed at natural analogues seek to quantify the rate and extent of CO₂ trapping via precipitation of carbonate mineral cements (i.e., 'carbon mineralization' or 'mineral-trapping'), and by so doing to make quantitative predictions about the probable long-term fate of CO₂ in a prospective reservoir. Stable carbon and oxygen isotope analyses are an important component of this line of research, as they provide a means by which to distinguish different carbonate cement populations within the pore space of a reservoir or caprock unit (e.g. CO₂-charge-related cements vs. those that may have formed during earlier stages of sediment alteration). However, clear interpretations of conventional isotopic data (sampling typically at the 100-1000 µm³ scale) will be hindered in situations where it is not possible to mechanically or chemically separate (due to small crystal size) different carbonate cement generations. This, specifically, is where in situ analysis by the SIMS technique has a unique advantage (routine in situ sampling at the 1-10 µm scale).

References Cited

Bonneville, A. et al., 2013, Evaluating the Suitability for CO₂ Storage at the FutureGen 2.0 Site, Morgan County, Illinois, USA: Energy Procedia, v. 37, p. 6125-6132, doi:10.1016/j.egypro.2013.06.541.

Bowen, B.B., R.I. Ochoa, N.D. Wilkens, J. Brophy, T.R. Lovell, N. Fischietto, C.R. Medina, and J.A. Rupp, 2011, Depositional and diagenetic variability within the Cambrian Mount Simon Sandstone: Implications for carbon dioxide sequestration: Environmental Geosciences, v. 18/2, p. 69-89, doi:10.1306/eg.07271010012.

Carroll, S.A., W.W. McNab, Z. Dai, and S.C. Torres, 2013, Reactivity of Mount Simon Sandstone and the Eau Claire Shale under CO₂ storage conditions: Environmental Science & Technology, v. 47/1, p. 252-261, doi:10.1021/es301269k.

Denny, A.C., R. Kozdon, K. Kitajima, M.G. Sliwinski, M. Spicuzza, and J.W. Valley, 2015, δ18O and δ13C-zoned carbonate cements as records of porewater temperature and composition in siliciclastics: In-Situ SIMS analyses from Early Paleozoic sandstones in the Illinois Basin, USA: AAPG Annual Convention and Exhibition.

Finley, R., 2005, An assessment of geological carbon sequestration options in the Illinois Basin: United States Department of Energy, 581 p.

Fishman, N.S., 1997, Basin-wide fluid movement in a Cambrian paleoaquifer: Evidence from the Mt. Simon sandstone, Illinois and Indiana, *in* I.P. Montanez, J.M. Gregg, and K.L. Shelton, eds., Basin-wide Diagenetic Patterns: Integrated Petrologic, Geochemical, and Hydrologic Considerations: Tulsa, Oklahoma, U.S.A.: SEPM, p. 221-234.

Hoholick, J.D., T. Metarko, and P.E. Potter, 1984, Regional variations of porosity and cement: St. Peter and Mount Simon sandstones in Illinois Basin: AAPG Bulletin, v. 68/6, p. 753-764.

Leetaru, H.E., S. Frailey, D. Morse, R.J. Finley, J.A. Rupp, J.A. Drahozval, and J.H. McBride, 2009, Carbon sequestration in the Mt. Simon Sandstone saline reservoir.

Leetaru, H.E., and J.T. Freiburg, 2014, Litho-facies and reservoir characterization of the Mt Simon Sandstone at the Illinois Basin - Decatur Project: Litho-facies and reservoir characterization of the Mt Simon Sandstone at the Illinois Basin: Greenhouse Gases: Science and Technology, v. 4/5, p. 580-595, doi:10.1002/ghg.1453.

Liu, F., P. Lu, C. Zhu, and Y. Xiao, 2011, Coupled reactive flow and transport modeling of CO₂ sequestration in the Mt. Simon sandstone formation, Midwest U.S.A.: International Journal of Greenhouse Gas Control, v. 5/2, p. 294-307, doi:10.1016/j.ijggc.2010.08.008.

Neufelder, R.J., B.B. Bowen, R.W. Lahann, and J.A. Rupp, 2012, Lithologic, mineralogical, and petrophysical characteristics of the Eau Claire Formation: Complexities of a carbon storage system seal: Environmental Geosciences, v. 19/3, p. 81-104, doi:10.1306/eg.02081211014.

Śliwiński, M.G., K. Kitajima, R. Kozdon, M.J. Spicuzza, J.H. Fournelle, A. Denny, and J.W. Valley, 2016a, Secondary ion mass spectrometry bias on isotope ratios in dolomite-ankerite, Part I: δ18O matrix effects: Geostandards and Geoanalytical Research, v. 40/2, p. 157-172, doi:10.1111/j.1751-908X.2015.00364.x.

Śliwiński, M.G., K. Kitajima, R. Kozdon, M.J. Spicuzza, J.H. Fournelle, A. Denny, and J.W. Valley, 2016b, Secondary ion mass spectrometry bias on isotope ratios in dolomite-ankerite, Part II: δ13C matrix effects: Geostandards and Geoanalytical Research, v. 40/2, p. 173-184, doi:10.1111/j.1751-908X.2015.00380.x.

Śliwiński, M.G., R. Kozdon, K. Kitajima, A. Denny, and J.W. Valley, 2016, Microanalysis of carbonate cement δ18O in a CO₂-storage system seal: Insights into the diagenetic history of the Eau Claire Formation (Upper Cambrian), Illinois Basin: AAPG Bulletin, v. 100/6, p. 1003-1031.

In-situ, Micron-scale Analyses of Carbonate δ^{13} C & δ^{18} O by SIMS: A Method for Identifying Carbonate Cements that Form in Response to CO, Injection at the Illinois Basin Decatur Project Site







M.G. Śliwiński^{1*}, K. Kitajima¹, R. Kozdon^{1,2}, M.J. Spicuzza¹, A. Denny¹ and J.W. Valley¹

¹ WiscSIMS, Department of Geoscience, University of Wisconsin-Madison, Madison, WI, 53706 (*msliwinski@wisc.edu) ² Lamont-Doherty Earth Observatory of Columbia University, Palisades, NY, 10964

Extended Abstract

mass spectrometry (SIMS) in carbon se-

sampling techniques in conventional isotope

ate δ^{18} O and δ^{13} C values.

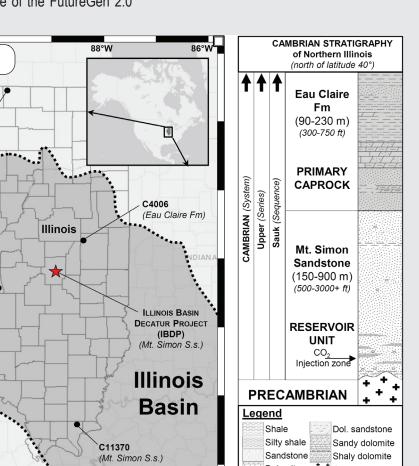
An important aspect of analytical met ods-advancement is the continued develop that fall along the dolomite-to-ankerite solic

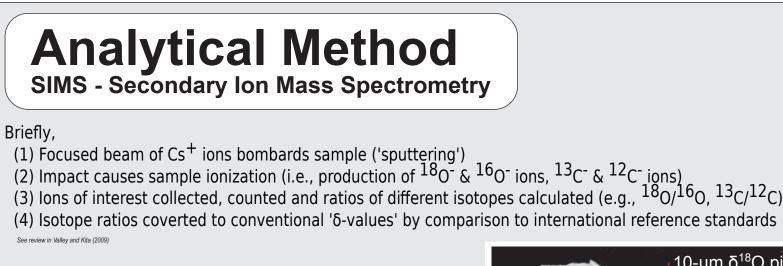
mated that at least 10-20% of the introduced CO, may be consumed, and thus permanently stored, via the precipitation Ca-Mg-Fe-carbonates (e.g., Liu et al. 2011 Carroll et al., 2013). We provide here a preliminary characterization of the pre-injection mineralogy and isotopic fingerprints $(\delta^{13}C \text{ and } \delta^{18}O) \text{ of carbonate cements in }$ the Mt. Simon Sandstone reservoir and the overlying shaly caprock (the Eau Claire Fm). By drawing upon published data on C-isotope composition of the injected CO₂, we also make simple predictions about the possible δ^{13} C signatures of different carbonate cement types that may form in response to imposed CO₂ storage.

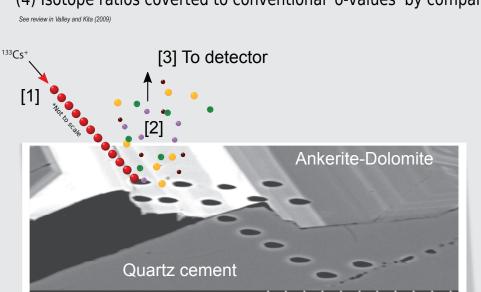
Carbonate cements are generally abundant (although somewhat heterogeneousl distributed) within the diverse suite of lithofacies that comprise the Eau Claire (e.g. Neufelder et al., 2012; Śliwiński et al., 2016) Cement mineralogy is predominantly dolo-

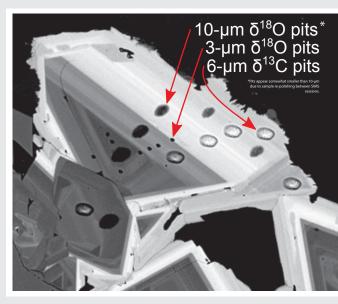
is situated within the lower Mt. Simon Sand-

lanned into the upper portion of this reservoir

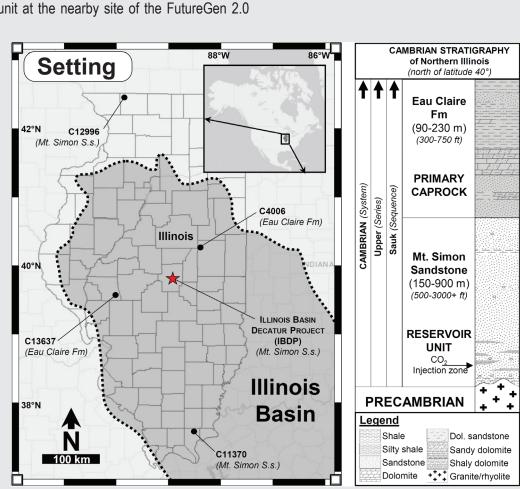




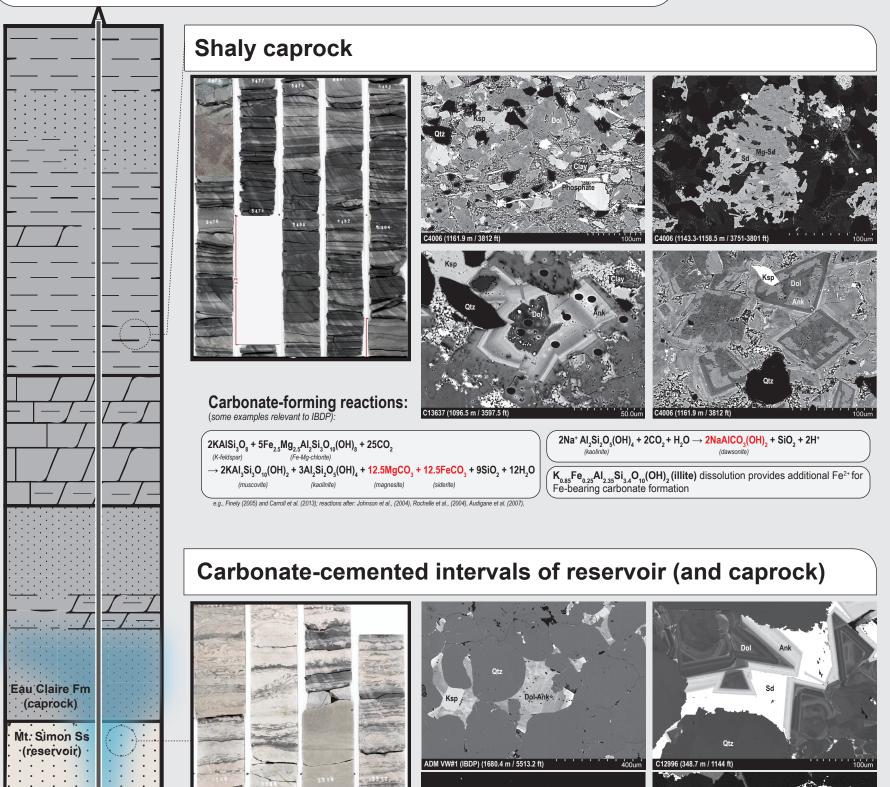


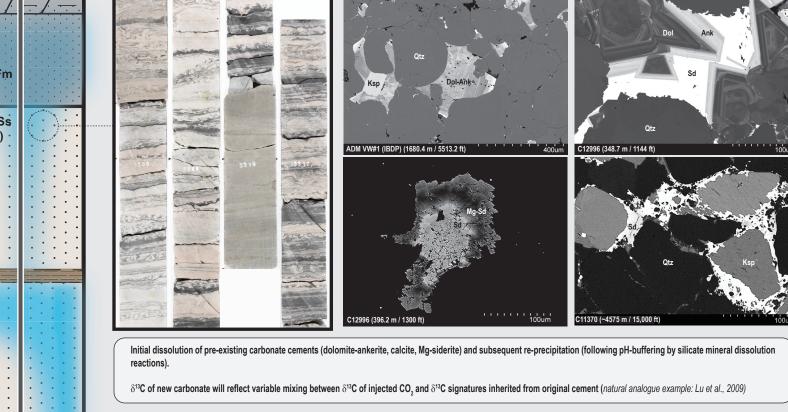


mite-ankerite, calcite and subordinate siderite project (now defunded; e.g., Bonneville et al.

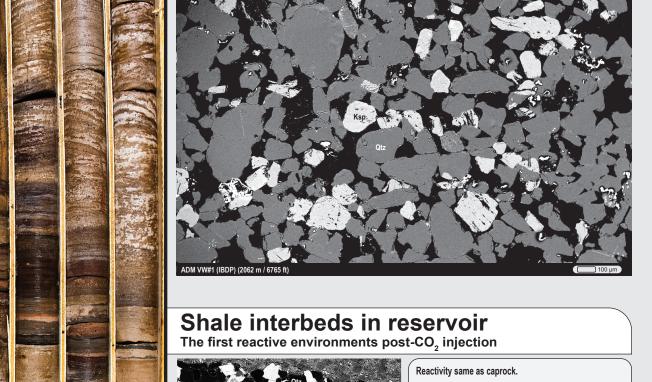


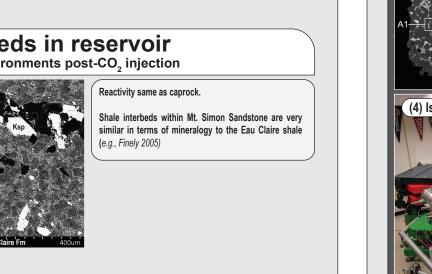
Reactive environments During CO, storage



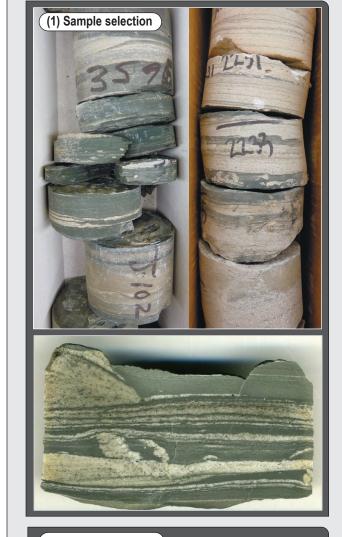


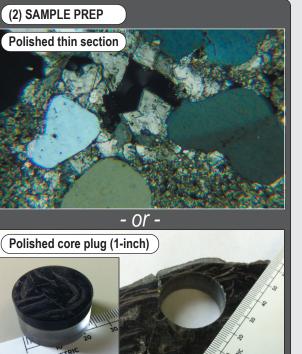
(Largely un-reactive)

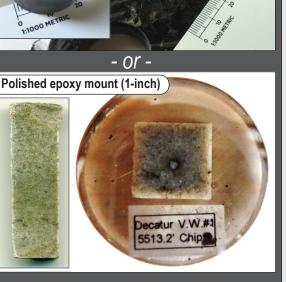




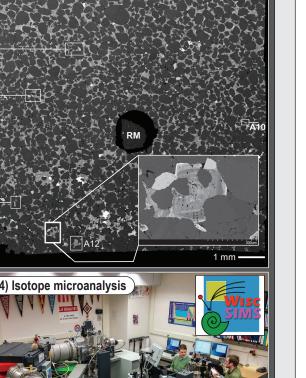
Process From core to crystal



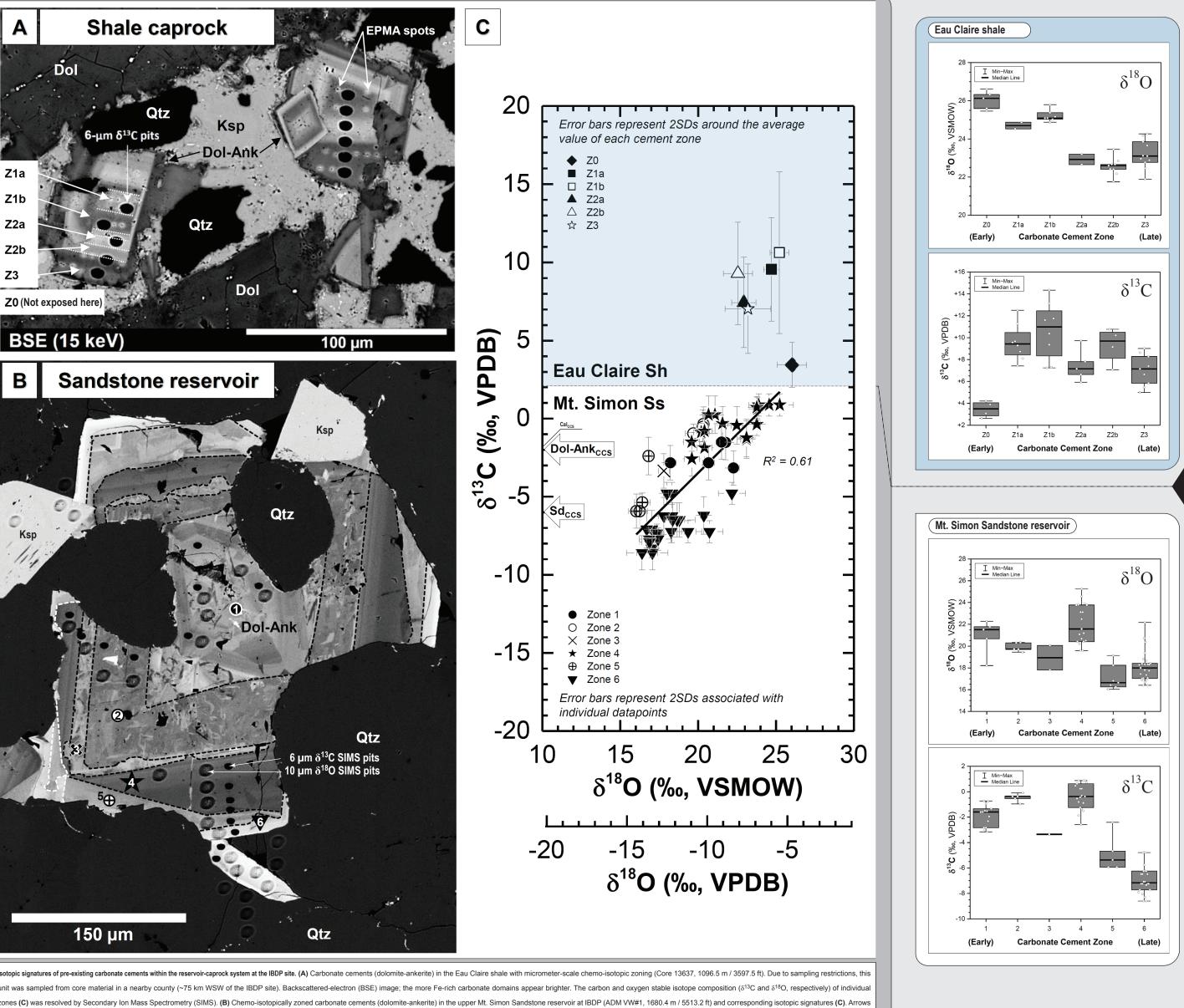


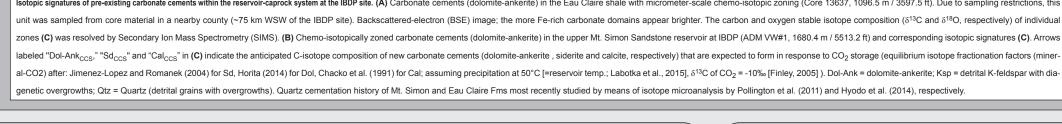


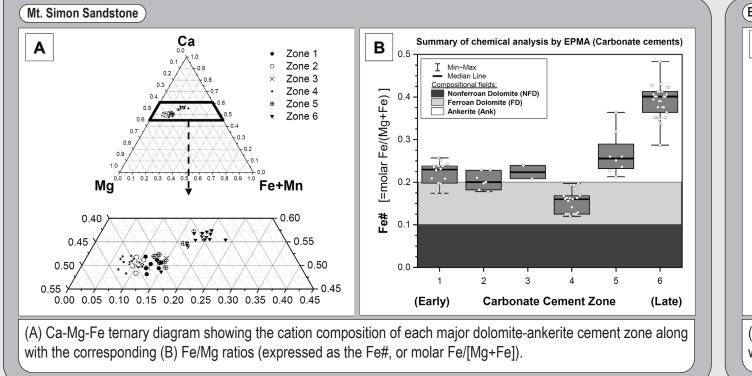
3) Region selection (petrography by SEM)

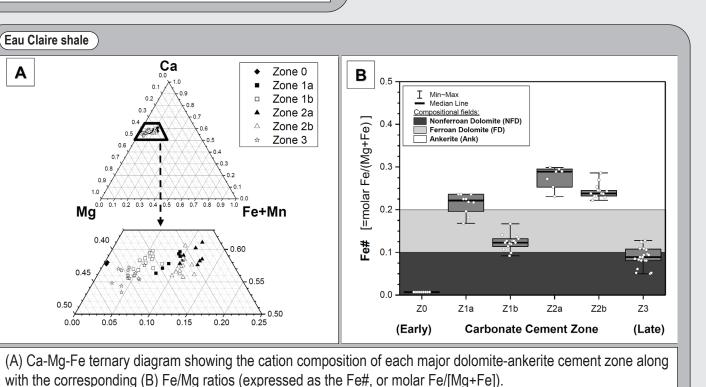


Pre-injection characterization Carbonate cements at the Illinois Basin Decatur Project reservoir-caprock system









Applications

The use of SIMS could provide unique insights to studies concerned with the following

- Verifying if, at what rate, and where in the reservoir-caprock system carbonate cements precipitate during engineered CO₂ storage
- Characterizing the isotopic composition of carbonate cements that pre-date the CO2-charge in engineered reservoirs or their natural analogues, especially in geological settings where such cements occur in abundance
- Assessing the impact of sequestration-related carbonate cements on pore-space geometry and pore-connectivity - a necessary first step is the ability to identify such cemens and to differentiate them from pre-existing carbonates
- Identifying associations between fracture-filling carbonate cements and CO, migration into / through the
- Determining or refining existing rate estimates (based) on bulk isotope analyses) of carbonate cement-forming reactions in natural analogue settings, with a focus on reservoir-caprock systems of contrasting mineralogical make-up; the aim of this is to better forecast the rate and extent of carbon mineralization (i.e., CO₂-trapping in mineral form) in the different types of prospective CO₂-storage reservoirs currently under consideration by the research community
- Analyzing the isotopic composition of experimental reaction products in instances where product volumes are sufficiently small so as to hinder analysis by conventional IRMS methods (this applies to laboratory-scale experiments designed to elucidate how CO₂ interacts with representative rock samples from a given reservoir-caprock system; e.g., Pollington et al.,
- Investigating the carbon source(s) tapped during carbon mineralization in prospective settings for geologic carbon storage other than deep saline aquifers. The insights gained from such investigations are essential for calibrating geochemical models/reservoir simulations that attempt to predict how the mineralogy of a given engineered reservoir-caprock system will evolve in the long-term in response to imposed CO₂ storage (and to derive more accurate estimates regarding the fraction of CO2 that will become securely trapped in mineral form over time)

This material is based primarily upon work supported by the US Department of Energy Office of Science, Office of Basic Energy Sciences, Chemical Sciences, Geosciences, and Biosciences Division undiaward number DE-FG02-93ER14389. The WiscSIMS Laboratory is partly funded by the US National Science Foundation (EAR-1355590). We thank our colleagues at the University of Wisconsin-Madison for providing instrumentation support: N. Kita (SIMS), J. Kern (SIMS), J. Fournelle (SEM & EPMA) and P. Gopon (SEM & EPMA). The Midwest Geological Sequestration Consortium is acknowledged for making samples from the Illinois Basin Decatur Project (IBDP) available for study and we thank A. Hyodo and the staff of the Illinois State Geological Survey for assistance with sampling.

, Kozdon, R., Kitajima, K., Denny, A. & Valley, J. W. Microanalysis of carbonate cement 8"0 in a CO₂-storage system seal: Insights into the diagenetic history of the Eau Claire Formation (Upper Cambrian), Illinois Basin. AAPG Bull. 100, 1003–1031 (2016). Kita, N. T. in Second. Ion Mass Spectrom. Earth Sci. Glean. Big Pict. Small Spot (ed. Fayek, M.) 41, 19–63 (Mineralogical Association of Canada (MAC), 2009).



(2) Impact causes sample ionization (i.e., production of $^{18}O^{-}$ & $^{16}O^{-}$ ions, $^{13}C^{-}$ & $^{12}C^{-}$ ion

