### CO<sub>2</sub> Gradient Affects the Microbial Ecology of a Potential Sequestration Site\*

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#### **Abstract**

Geological carbon sequestration is likely to be part of a comprehensive strategy to minimize the release of greenhouse gasses into the atmosphere. Reservoir storage capacities and long-term security of geologic carbon sequestration will be dependent on the trapping mechanisms and mineral transformation in the deep subsurface. A critical need exists to understand the evolution of microbial populations that influence the biogeochemistry in these reservoirs. As the CO<sub>2</sub> front moves through the storage aquifer, microbial communities may preside in residual brine left behind in cracks, dead flow zones, and upstream to the front; this brine will have a gradient of dissolved CO<sub>2</sub> concentration.

The evolution of microbial ecology along this CO<sub>2</sub> concentration gradient was investigated using fluid-slurry samples obtained from a proposed carbon sequestration site. The native species of these samples were investigated with varying pCO<sub>2</sub> from 0% to 100% under reservoir temperature and pressure for up to 56 days. Microbial growth occurred with 0% and 1% pCO<sub>2</sub>, while microbial population decreased by four orders of magnitude with 10% and 100% pCO<sub>2</sub>. This suggests that any biological processes that may have been involved in the security of the reservoir pre-CO<sub>2</sub>-injection will be hindered near the CO<sub>2</sub> plume, but may be important where CO<sub>2</sub> is attenuated. Batch reactors with 0% pCO<sub>2</sub>, but lowered pH had increased microbial population by one order of magnitude, suggesting that microbial shifts will occur even in highly buffered reservoirs due to exposure to CO<sub>2</sub>. Halotolerant microorganisms Halomonas and Marinobacter were the most tolerant to the conditions that would follow CO<sub>2</sub> injection. Findings provide insight into the populations that may survive in the deep subsurface following the supercritical fronts. These populations will eventually give rise to the community that will impact the trapping mechanisms, storage capacities, and long-term security of the CO<sub>2</sub> deposits. Knowledge of the surviving microbial populations will enable improved models for predicting the fate of CO<sub>2</sub> following injection and lead to better strategies for ensuring the security of CO<sub>2</sub> in the subsurface.

#### **Reference Cited**

Rackley, S.A., 2010, Carbon capture and storage: Butterworth-Heinemann/Elsevier, Burlington, MA, 392 p.

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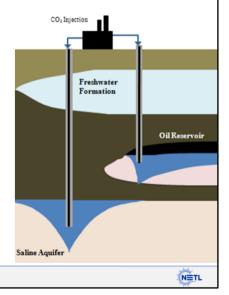
# CO<sub>2</sub> gradient affects the microbial ecology of a potential sequestration site Djuna Gulliver, Greg Lowry, Kelvin Gregory National Energy Technology Laboratory and Carnegie Mellon University Pittsburgh, PA

### **Background**

Geologic carbon storage

#### · Main carbon storage concerns:

- capacity of CO<sub>2</sub> storage
- permanence of CO<sub>2</sub> storage
- reservoir injectivity



Presenter's notes: Just hit 400 ppm in atmosphere, compared to 368 in 2000 (Rackley, 2010).

# Background Biological processes

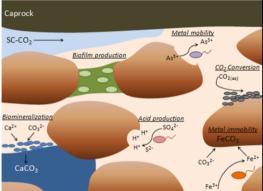
Microbiology may affect geologic CO<sub>2</sub> storage through various

biological processes

- Biomineralization

Biofilm production

- Acid production
- Metal mobilization
- Metal immobilization
- CO<sub>2</sub> conversion



 We need to understand the relevant microbial community before we can begin to investigate bioprocesses that will occur.

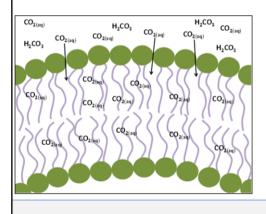


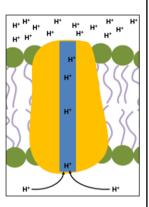
Presenter's notes: On a large these biological processes will make an impact. Numerous instances in engineered and natural systems. Use numerous bioremediation sites around the world, engineers have been able to control ecology. Concerned about what we do intentionally and unintentionally.

### **Background**

CO<sub>2</sub> effects on microbiology

- Affect cytoplasmic enzymes
- Greatly lowers pH
- Membrane infiltration of CO<sub>2</sub>





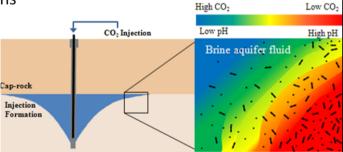


# Background Heterogeneous characterization

• Injection of CO<sub>2</sub> will not lead to a uniform CO<sub>2</sub> concentration

Varying conditions

- CO<sub>2</sub> conc.
- pH
- Ion conc.



Microbial ecology will adjust to varying condition



### **Objectives**

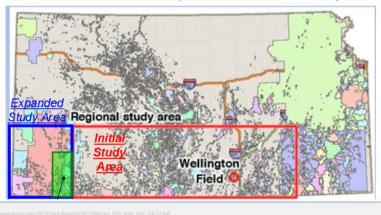
- Characterize the native microbial community of the Arbuckle Aquifer, Wellington, KS
- Characterize Arbuckle Aquifer microbial community changes with various CO<sub>2</sub> concentration



# Materials and Methods Sampling Site

Depth of 4175-4390 ft

Formation consists of dolomite with lenses of shale Pressure of 14 MPa, Temperature of 40 °C, pH of 7.8

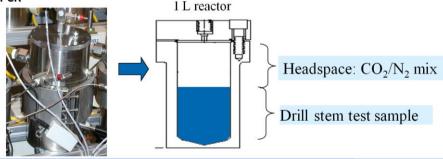


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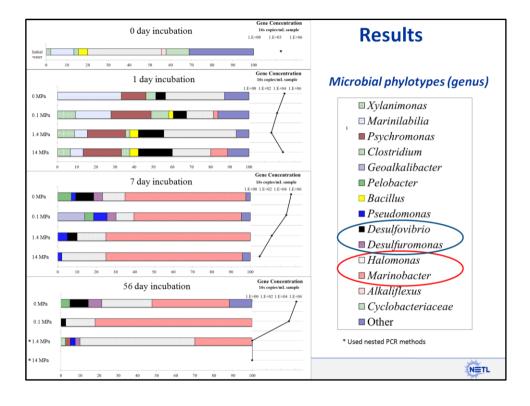
#### **Materials and Methods**

#### Pressurized vessel

- Place drill stem test samples under in situ temperature (40 °C) and total pressure (14 MPa)
- Vary partial pressure of CO<sub>2</sub> of 0 MPa pCO<sub>2</sub>, 0.1 MPa pCO<sub>2</sub>, 1.4 MPa pCO<sub>2</sub>, and 14 MPa pCO<sub>2</sub>
- Vary incubation time of 1 day, 7 days, and 56 days to detect adapting ecology
- Characterized change in microbial community with 16S rRNA clone libraries and qPCR



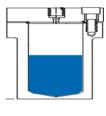




## **Materials and Methods**

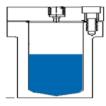
**Pressurized vessel** 

# Vessel 1



0 MPa  $CO_2$ , pH=4.4

# Vessel 2



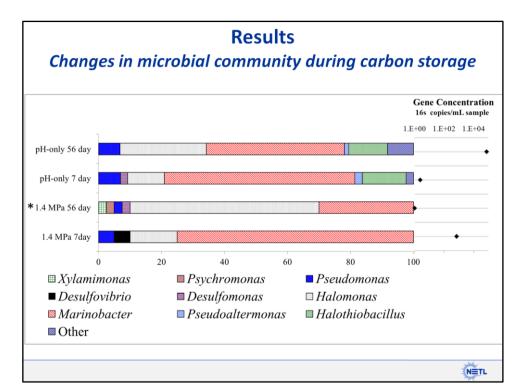
1.4 MPa CO<sub>2</sub>

Liquid: Arbuckle Drill Stem Test

Temperature: 40 °C

Run time: 7 days, 56 days





### **Summary**

### Changes in microbial community during carbon storage

- CO<sub>2</sub> exposure as low as 0.1 MPa will inhibit microbial growth
- CO<sub>2</sub> exposure as low as 1.4 MPa will exterminate microbial growth
- Highly diverse samples after 1 day evolved to limited diversity after 7 days and 56 days of CO<sub>2</sub> exposure
- Halotolerant microorganisms Halomonas and Marinobacter appeared to be the most resilient to CO<sub>2</sub>
- Microbial processes will not be of importance near the CO<sub>2</sub> front, but may be important at a distance where CO<sub>2</sub> will be attenuated



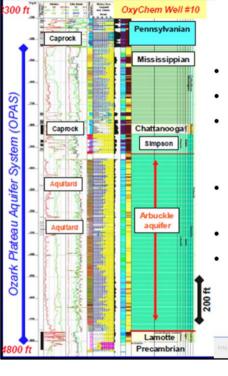
### **Acknowledgements**











# Additional Information Site Description

- Depth of 3,500 ft
- Formation thickness of 800 ft
- Overlain by 50 ft Chattanooga shale and Mississippian EOR reservoir
- Formation consists of dolomite with shaley aquitards
- Pressure of 14 MPa
- Temperature of 40 °C

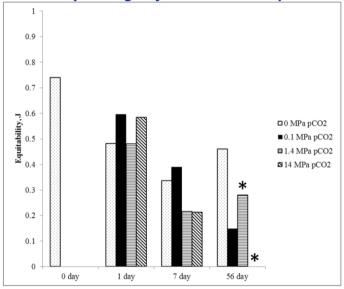
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# Additional Information Site Description

- DOC of 110-170 ppm C
- CH<sub>4</sub> of 280-4370 ppm
- CO<sub>2</sub> of 0.40-4.0 ppm
- TDS of 10,000-250,000 ppm
- Salinity of 55,100-57,800 ppm

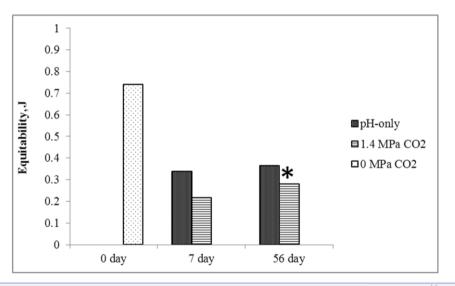


# Additional Information Diversity change of main vessel experiment











# Additional Information Water Chemistry

	Units	Raw	0 MPa CO <sub>2</sub>	0.1 MPa CO <sub>2</sub>	1.4 MPa CO <sub>2</sub>	14 MPa CO <sub>2</sub>
Ca	mg/L	2502.0	2071.0	2186.0	2186.0	2104.0
Fe	mg/L	$< D\Gamma$	$< D\Gamma$	< DL	0.5	2.5
K	mg/L	258.5	251.7	255.9	253.2	256.6
Mg	mg/L	702.4	567.8	578.4	577.1	590.7
Mn	mg/L	0.9	0.7	0.9	0.9	0.9
Na	mg/L	22690.0	18280.0	18440.0	17530.0	17640.0
P	mg/L	1.5	3.4	1.4	2.4	2.0
S	mg/L	631.0	593.9	618.4	601.0	613.1
Si	mg/L	10.0	9.4	10.1	10.8	11.0
Zn	mg/L	0.2	0.1	0.8	1.8	2.4
SO42-	mg/L	1190.0	369.9	1007.4	953.7	1096.3
Cl	mg/L	31528.2	31992.3	32397.2	31639.1	37757.3
Modeled pH		7.80	7.50	5.50	4.38	2.88

