

# **Chemical Flooding the Lansing-Kansas City Formation in Kansas\***

**Mark Ballard<sup>1</sup>**

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<sup>1</sup>Tertiary Oil Recovery Program (TORP), The University of Kansas, Lawrence, KS ([markt@ku.edu](mailto:markt@ku.edu))

## **Abstract**

The Tertiary Oil Recovery Program at the University of Kansas in cooperation with the Research Partnership to Secure Energy for America (US DOE funded) and industry partners are presently designing a chemical flood for implementation in the Trembley Oilfield, Reno County, KS. The purpose of the project is to test and demonstrate the performance of chemical flooding. An overview of chemical flooding will be presented that will include the basics of the oil-recovery process and considerations for where it might be applied. A summary of the field project and the progress to-date will be reported. The chemical flood design covers laboratory testing to formulate a chemical system that achieves desired performance for the Trembley reservoir (oil, water and rock) and evaluation of the reservoir to design the field implementation. Results of both the laboratory work and the field evaluation will be presented. The Trembley Oilfield produces from a thin bed of oolitic grainstone in the Pennsylvanian Lansing-Kansas City (LKC) interval. Oil production was initially by fluid expansion and like many LKC fields, it has been successfully waterflooded. The Trembley has favorable characteristics to be chemical flooded and good performance should lend promise to the application of chemical flooding of other LKC reservoirs.

## **References Cited**

Chatzis, I., and N.R. Morrow, 1984, Correlation of capillary number relationships for sandstone: SPE Journal, v. 24/5, p. 555-562.

Huh, C., 1979, Interfacial tensions and solubilizing ability of a microemulsion phase that coexists with oil and brine: J. Colloid Interface Science, v. 71, p. 408-426.

# AAPG Mid-Continent Section Meeting

Chemical Flooding the Lansing-  
Kansas City Formation in Kansas



Mark Ballard, Petroleum Engineer

# Our Partners

Funding



Field Partner



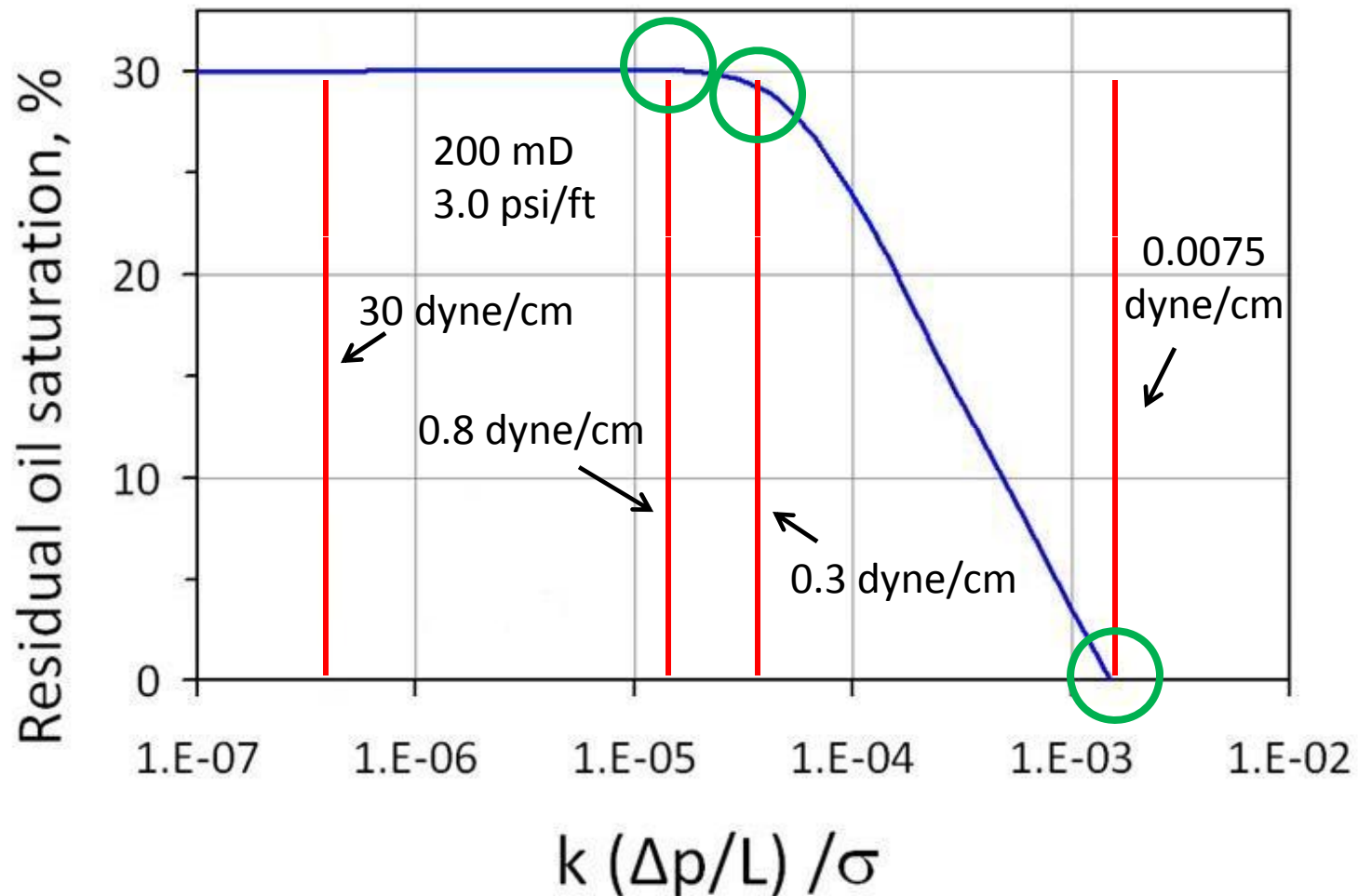
Surfactant and Formulation



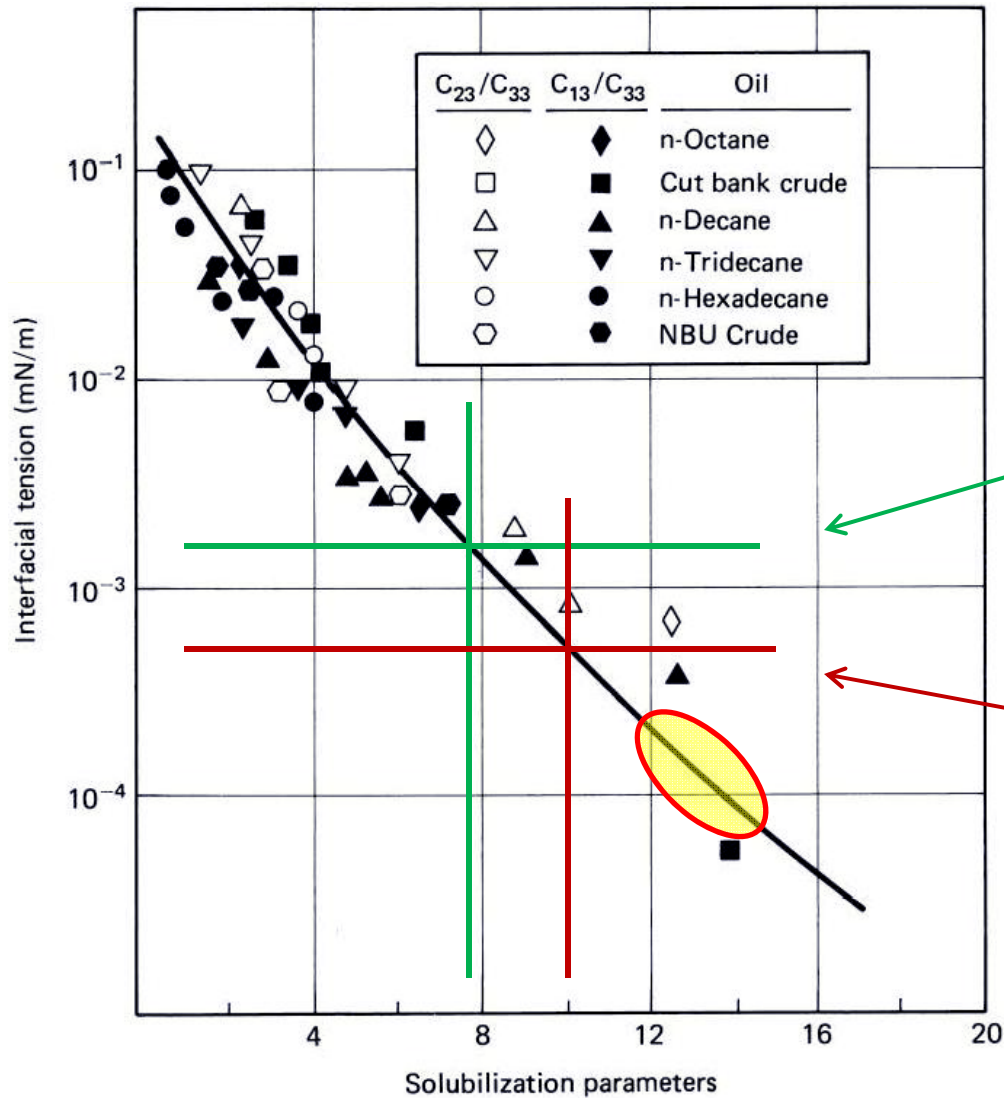
Polymer and field equipment



# Capillary Number



# Solubilization Ratio => IFT



$$\text{IFT} = \frac{\text{Salinity}}{\text{Solubilization}^2}$$

IFT = 0.0075

Solubilization Ratio > 10

Chun Huh, 1979

# What does it take?

- Field (target) selection
- Chemical system
- Field implementation

# Reservoir Selection

- Field (target) selection
  - Connected Flow Units
  - Volumetric Sweep

# Reservoir Selection

## *Responsive waterflood*

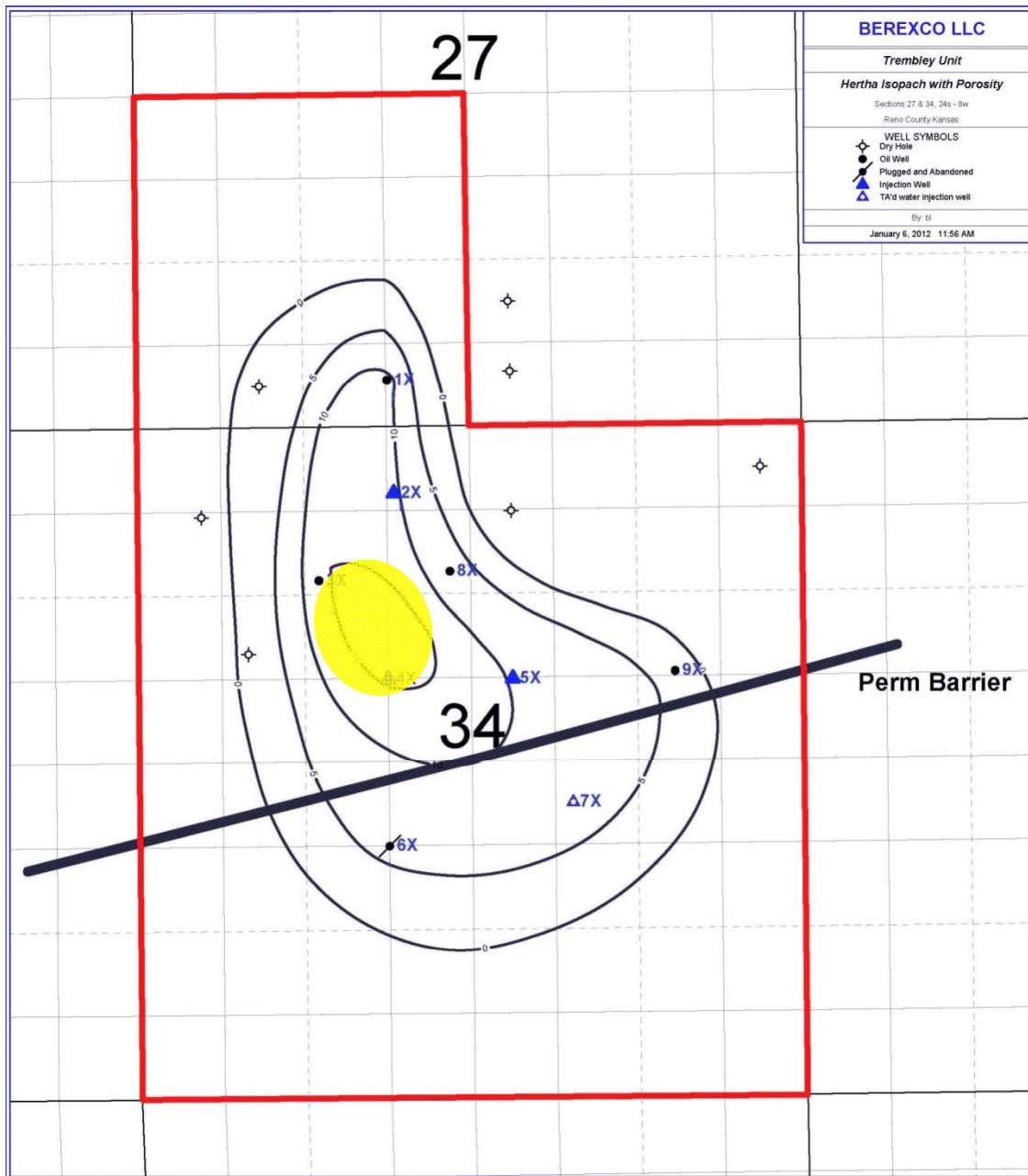
Floodable

Significant reservoir sweep

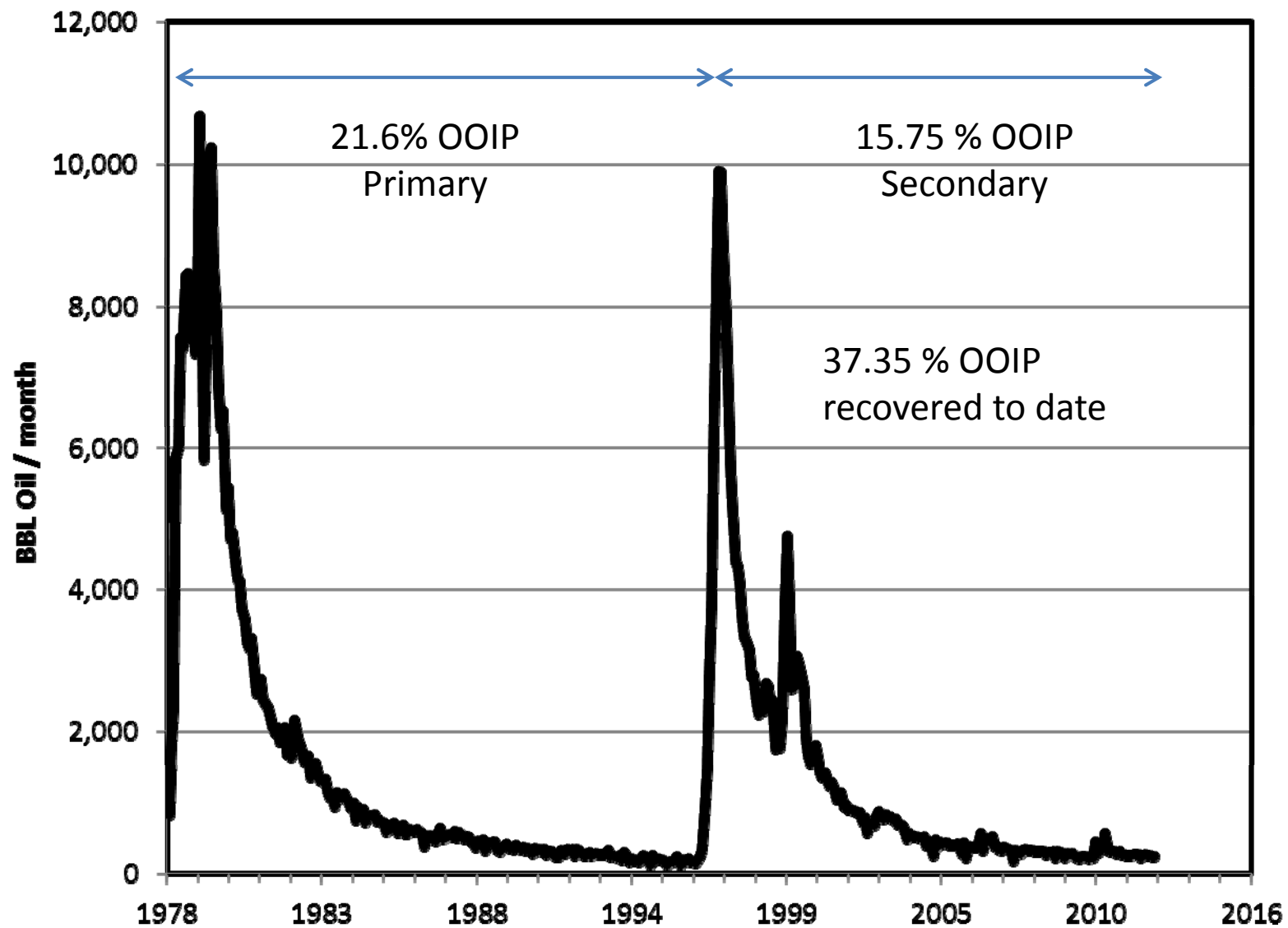
More reservoir data

Available surface facilities

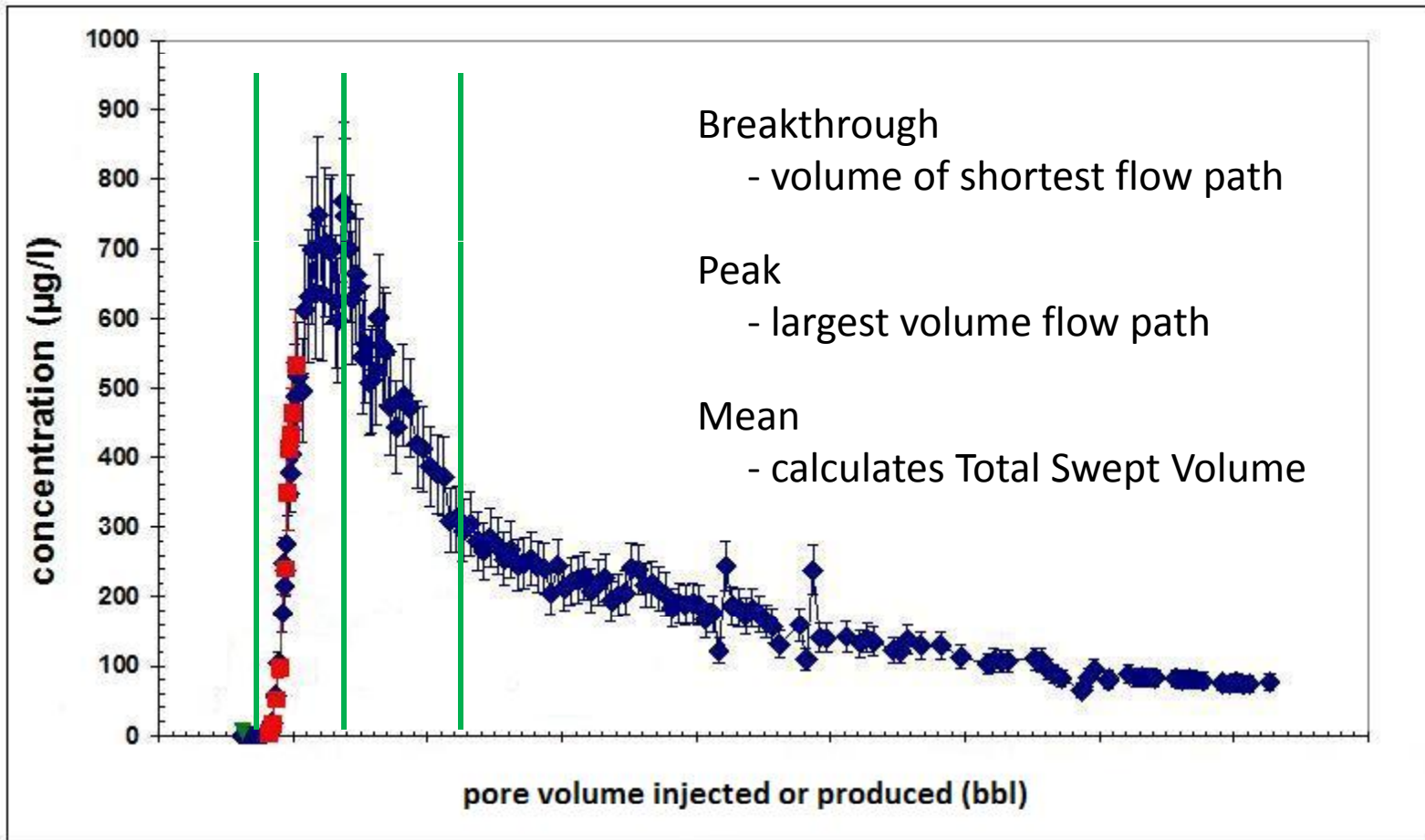




## Trembley field



# Tracers – connected flow units



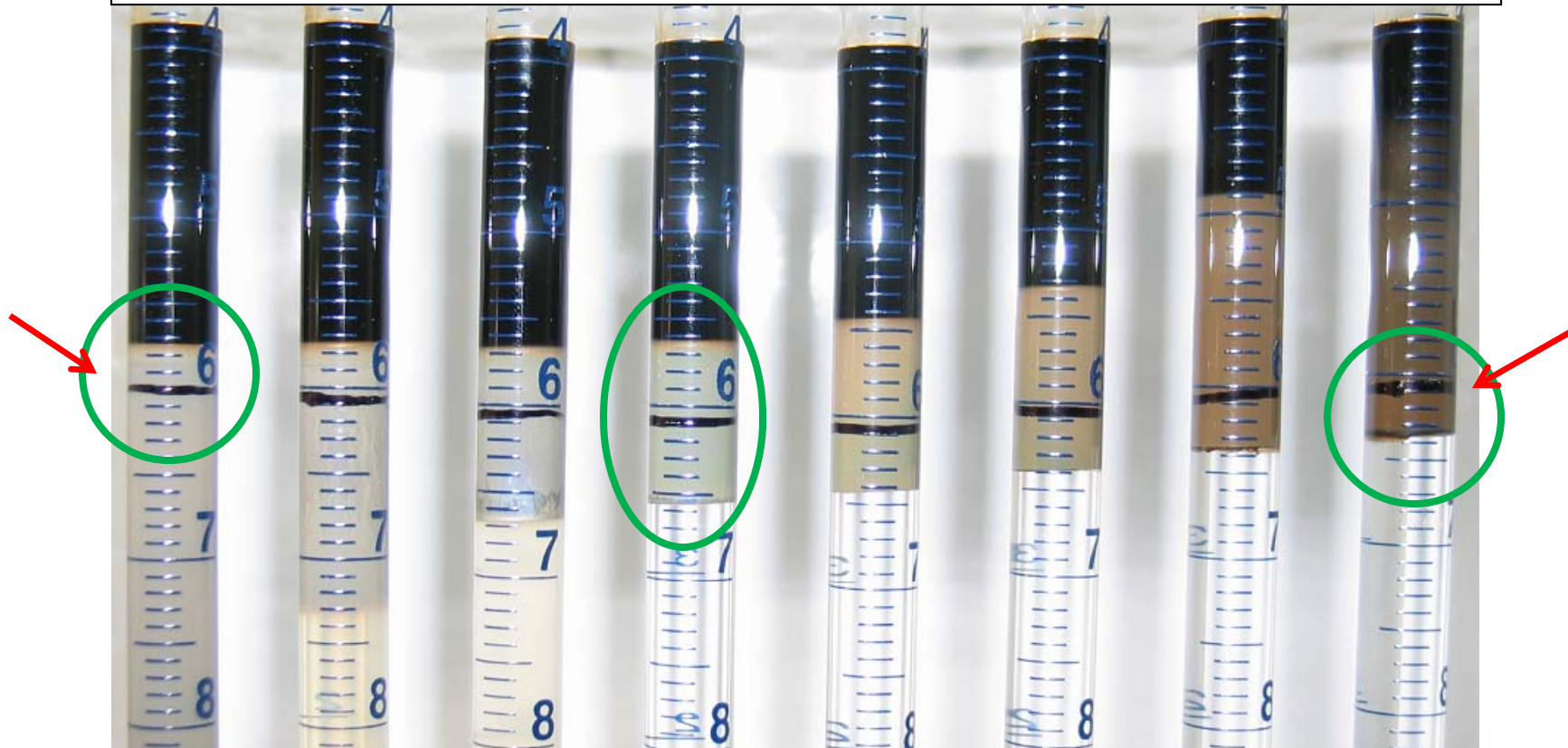
# Chemical System - Formulation

- Several variables
  - **Solubilization** (IFT)
  - Salinity
  - WOR
  - Aqueous Phase Stability (APSL)
  - **Adsorption**

# Chemicals

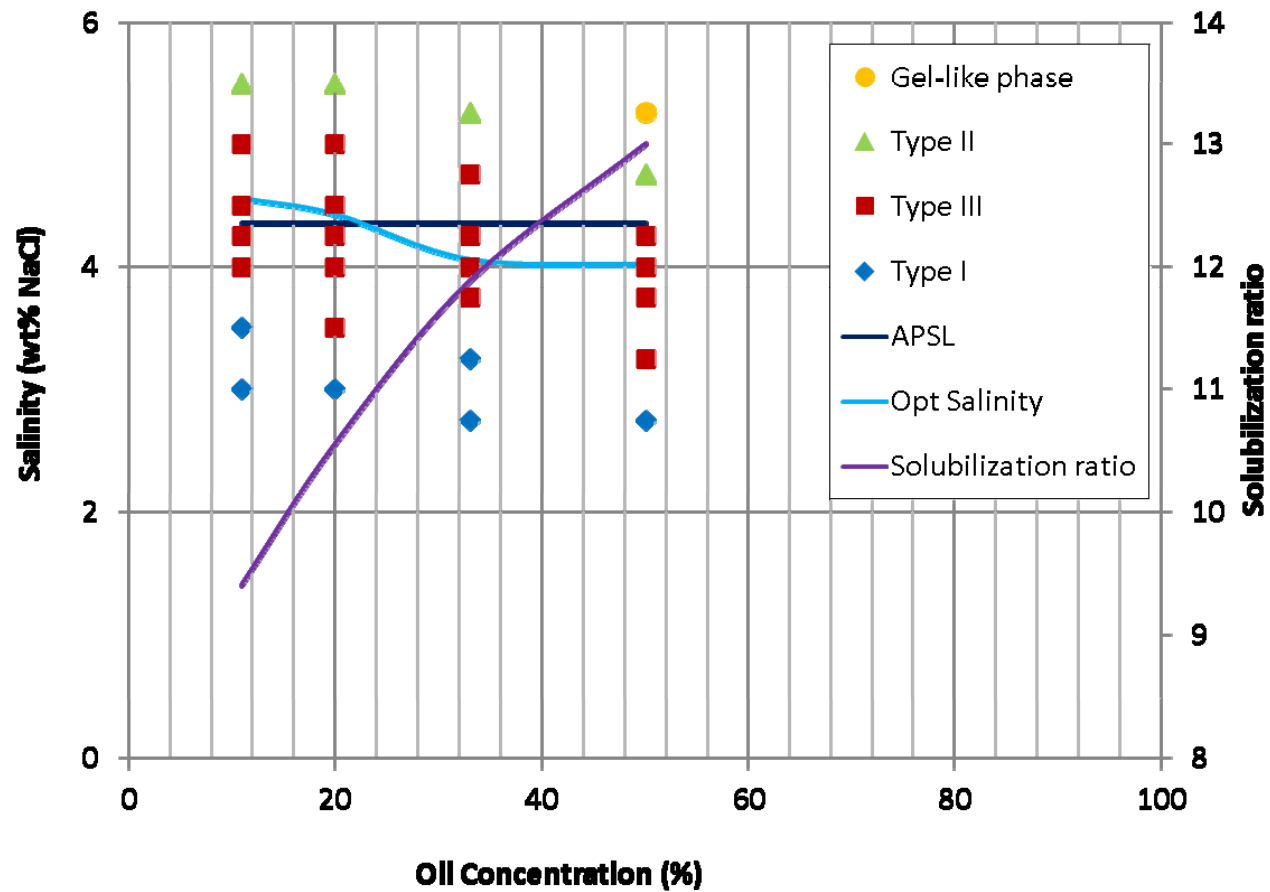
- **Surfactants** - mobilize trapped oil
- Solvents - enhance solubility
- Sacrificial agents - reduce adsorption
- Salts - affects behavior
- **Polymer** - min. mixing /max. sweep
- Water

NaCl conc (%)							
3.25	3.50	3.75	4.00	4.25	4.50	4.75	5.00

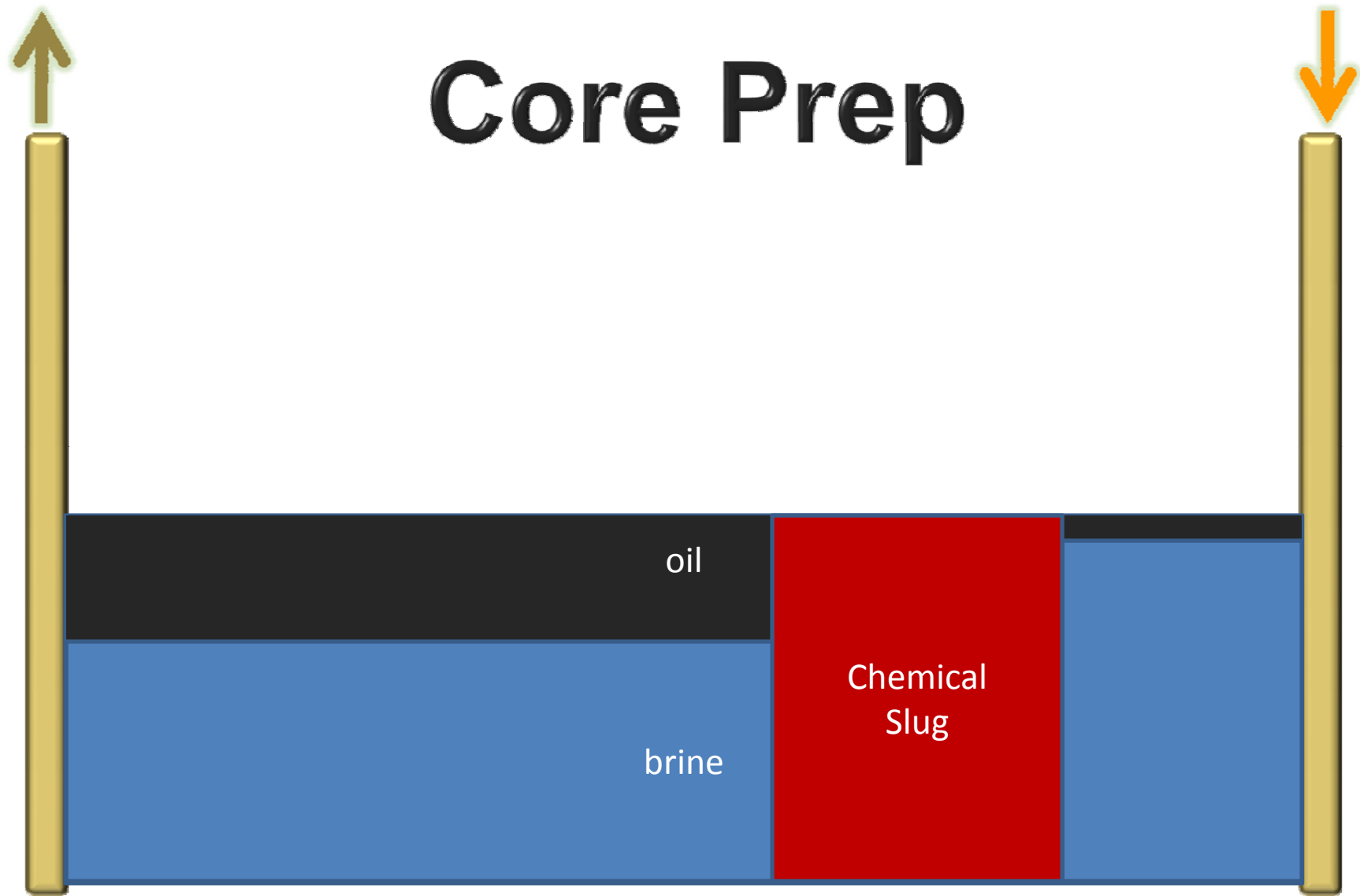


0.5% XOF-100S      2.0% NaCO<sub>3</sub>  
 0.1% XOF-600S      0.2% SNF 3330S

# Activity Diagram



# Core Prep

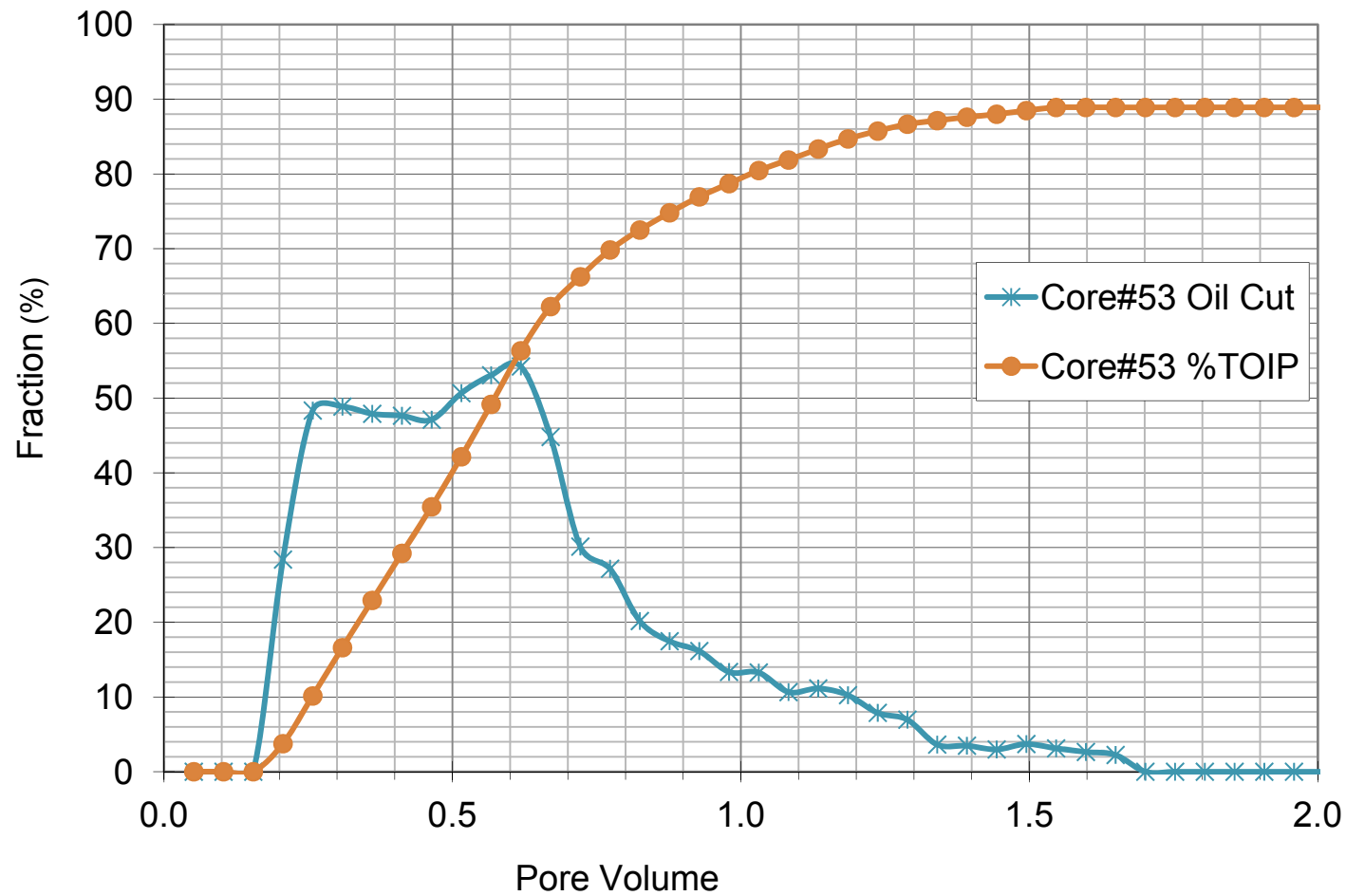




(Berea)

# Core 53

(includes alkali)



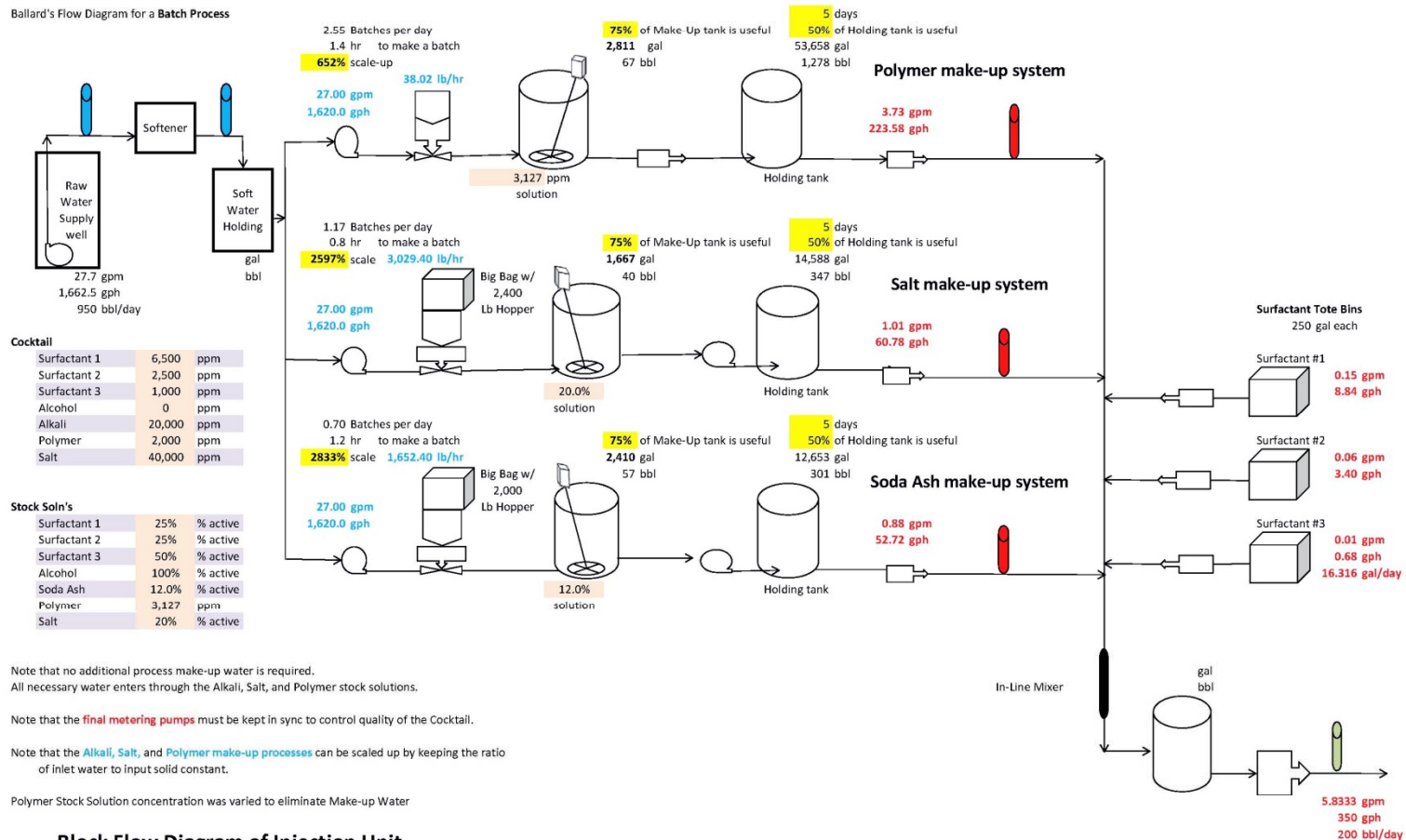
100R10, A10, P20, L42-20

# What does it take?

- Field (target) selection
- Chemical system
- Field implementation
  - Injectivity
    - Well workovers
  - Chemical Slug make-up
    - Quality control !!!
  - Production well testing
    - Slug transit & breakthrough

# Injection Plant

Ballard's Flow Diagram for a Batch Process



Block Flow Diagram of Injection Unit

# Injection Plant

Instructions: Yellow boxes require data entry

- (1) Enter injection rates
- (2) Enter wellhead or pump outlet maximum pressure
- (3) Enter the minimum polymer maturation time
- (4) Enter the ASP formula
- (5) Enter data for the state of the delivered chemicals
- (6) Check the Stock Solutions for accuracy
- (7) Enter the actual size of the Polymer Maturation Tank

(1) Injection Rate	200	bbl/day
(2) Max Pressure	1,200	psi
(3) Min. polymer maturation time	60	min

Notes:

The Cocktail density is IMPORTANT. Many subsequent calculations are based upon it.  
Most calculations are based upon the "Consumption, lb/day"  
Densities of constituents delivered as liquid are also important to get the Cocktail right  
Densities of Stock Solutions are important to get the Cocktail right

1.0 wt% = 10,000 ppm

(4) ASP Formula		
Surfactant 1	6,500	ppm
Surfactant 2	2,500	ppm
Surfactant 3	1,000	ppm
Alcohol	0	ppm
Alkali	20,000	ppm
Polymer	2,000	ppm
Salt	40,000	ppm

(5) Chemical Delivered State					
Phase/Form	Activity	Size	Type	Density	
Liquid	25%	250	gal tote	8.58	lbs / gal
Liquid	25%	250	gal tote	8.58	lbs / gal
Liquid	50%	250	gal tote	8.58	lbs / gal
Liquid	100%	250	gal tote	6.75	lbs / gal
Powder	100%	2,000	lb super-sack	60.00	lb / ft <sup>3</sup>
Powder	100%	55	lb sack	49.92	lb / ft <sup>3</sup>
Powder	100%	2,400	lb super-sack	79.00	lb / ft <sup>3</sup>

(6) Stock Solutions:				Density	
Surfactant 1	25%	% active		8.58	lbs / gal
Surfactant 2	25%	% active		8.58	lbs / gal
Surfactant 3	50%	% active		8.58	lbs / gal
Alcohol	100%	% active		6.75	lbs / gal
Soda Ash	12.0%	% active		9.22	lbs / gal
Polymer	3127.5	ppm		8.34	lbs / gal
Salt	20%	% active		9.60	lbs / gal
Cocktail				8.333	lbs / gal

Stock Solution Metering Pumps			
gal/min	gal/hr	gal/day	bbl/day

Consumption (as delivered)					
lb/hr	lb/day	lb/mo	sack/tote per mo	gal/day	gal/mo

Polymer Maturation tank			Main Inj Pump	
gal	bbl	ft <sup>3</sup>	HP	kW

Metering Pumps

Surfactant 1 (stock soln)	0.15	8.84	212.1	5.05	75.83	1,820	55,508	25.9	212.1	6,469
Surfactant 2 (stock soln)	0.06	3.40	81.6	1.94	29.17	700	21,349	10.0	81.6	2,488
Surfactant 3 (stock soln)	0.01	0.68	16.3	0.39	5.83	140	4,270	2.0	16.3	498
Alcohol	0.00	0.00	0.0	0.00	0.00	0	0	0.0	0.0	0
Alkali (stock solution)	0.88	52.72	1,265	30.13	58.33	1,400	42,698	21.3		
Polymer (stock soln)	3.73	223.58	5,366	127.76	5.83	140	4,270	77.6		
Salt (stock solution)	1.01	60.78	1,458.8	34.73	116.66	2,800	85,397	35.6		

Dilution Water Rate	0.00	0.00	0.00	0.00
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Main injection pump	5.83333	350	8400	200
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4.10	3.05
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Polymer Maturation Tank

Minimum Maturation Tank Working Volume

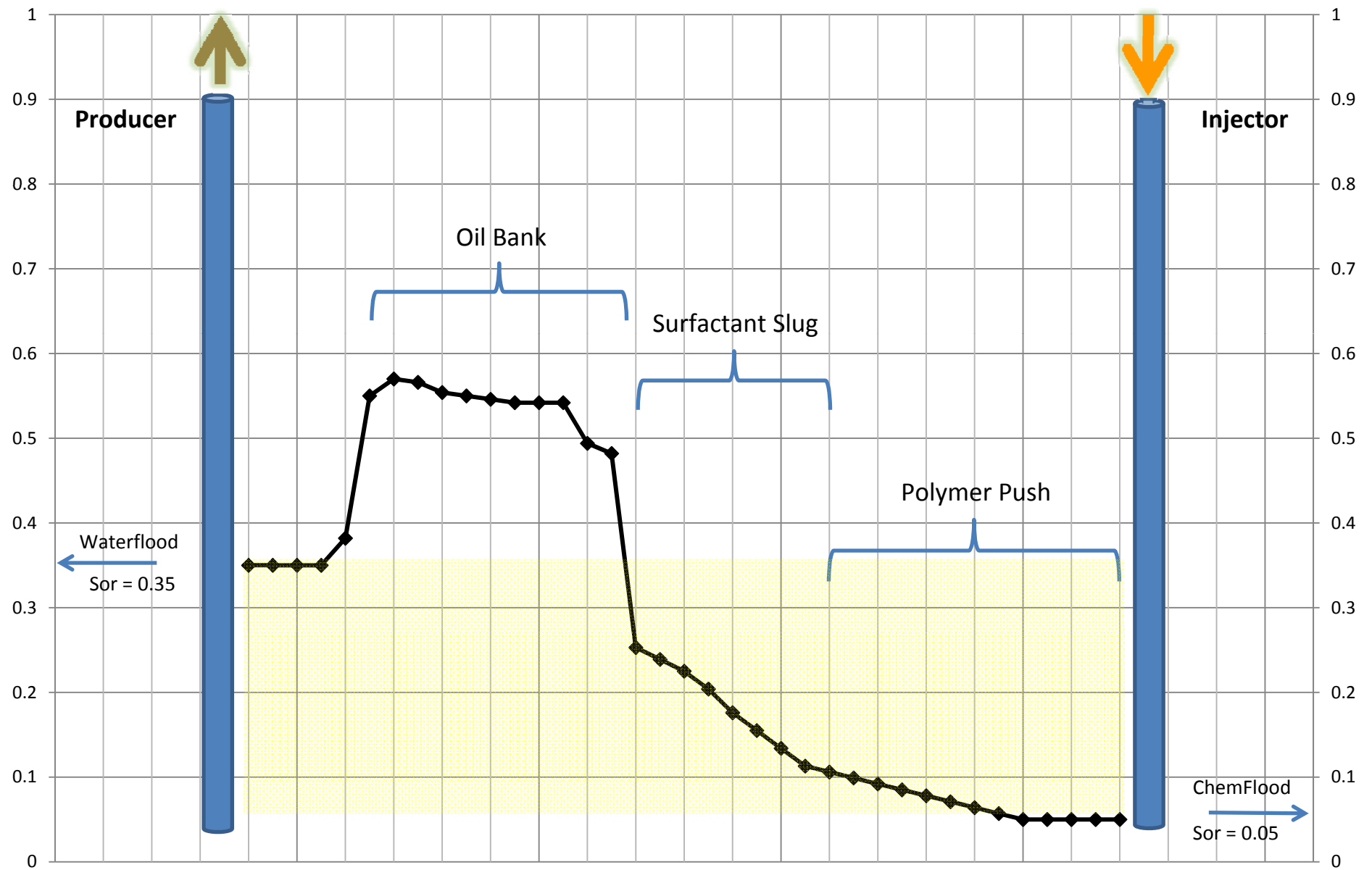
Recommended Total Tank Volume

(7) Actual Tank size specified

Actual polymer maturation time

223.58	5.32	29.89
279.47	6.65	37.36
279.47	6.65	37.36
75.00 minutes		
1.25 hours		

## Oil Saturation Profile between Injector and Producer



PV	0.052	0.103	0.155	0.207	0.258	0.310	0.362	0.413	0.465	0.517
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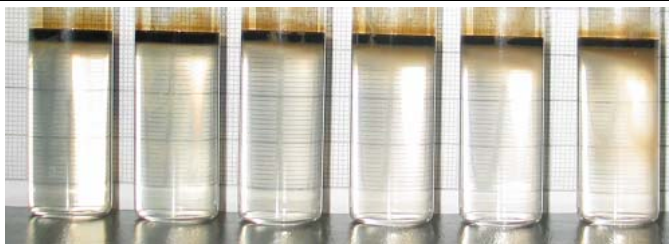
PV	0.568	0.620	0.672	0.723	0.775	0.827	0.878	0.930	0.981	1.033
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PV	1.085	1.136	1.188	1.240	1.291	1.343	1.395	1.446	1.498	1.550
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PV	1.601	1.653	1.705	1.756	1.808	1.860				
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Core 53  
effluent  
collection  
vials

# Concluding Remarks

- Great performance in the lab
- Translating performance to field is challenging
- Advantages for success
  - good waterflood performance
  - inter-well tracer study
  - core material
  - good field data
- **Successful demonstration in the Trembley**



# Thank You

Funding



Field Partner



Surfactant and Formulation



Polymer and field equipment

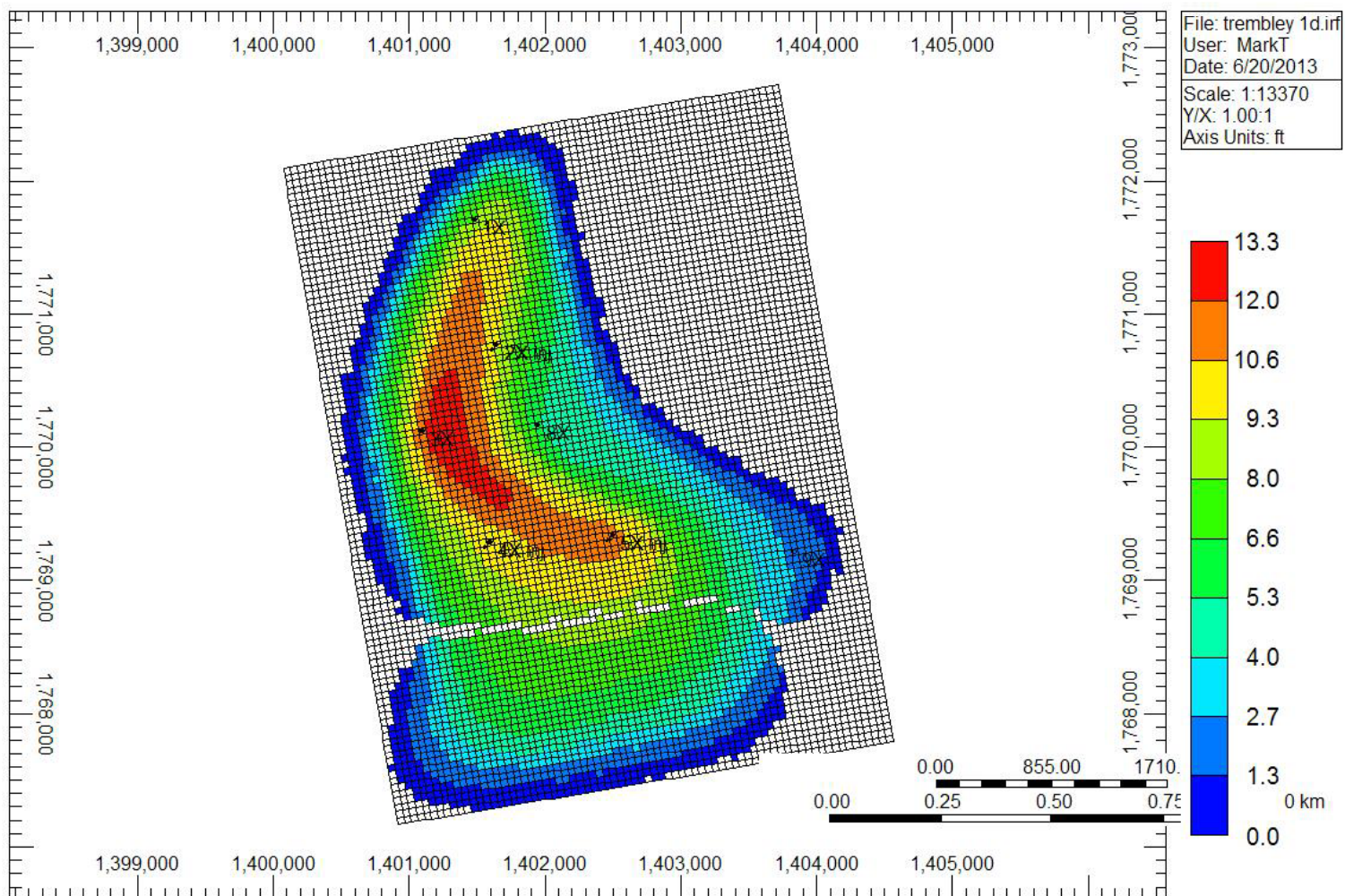






# Simulation

Grid Thickness (ft) 2013-01-01 K layer: 1



# Simulation

